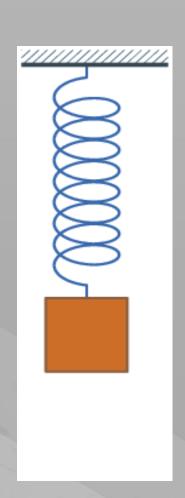
Building Materials

Lecture 4

Mechanical properties

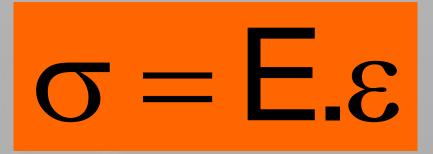


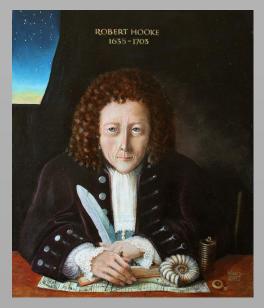
Modulus of elasticity



Elastic behavior of materials describes

Hooke's Law:





Robert **Hooke** 1635-1703

- ε ... strain [unitless]
- σ... stress [MPa]
- E ... modulus of elasticity [MPa] (Young's modulus)

Elastic modulus

$$\mathbf{E} = \frac{\mathbf{\sigma}}{\mathbf{\epsilon}_{\mathsf{el}}}$$

- the mathematical description of a material's tendency to be deformed elastically when a force is applied to it
- Hooke's law is valid only for elastic range of material

Elastic modulus

tension or compression - Young's modulus

$$\mathbf{E} = \frac{\mathbf{\sigma}}{\varepsilon_{\mathsf{el}}}$$

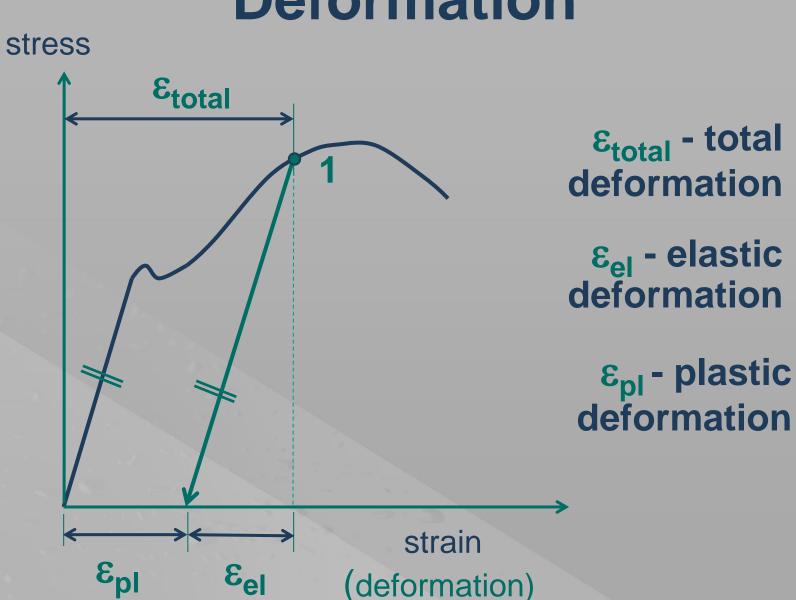


Thomas **Young** (1773-1829)

shear - modulus of rigidity or shear modulus

$$G = \frac{\tau}{\gamma}$$

Deformation

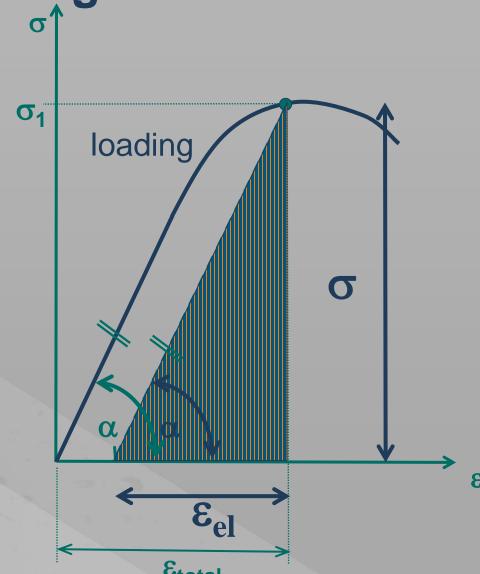


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Graphical determination of Young's Modulus

$$\mathsf{E} = \frac{\sigma}{\epsilon_{\mathsf{el}}}$$

$$E = tg \, \alpha$$



Young's modulus determination

statical

$$\mathbf{E} = \frac{\mathbf{\sigma}}{\varepsilon_{\mathsf{el}}}$$

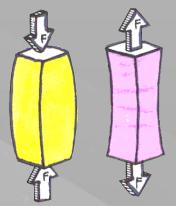
stress [MPa]

ε strain [-]

Young's modulus determination

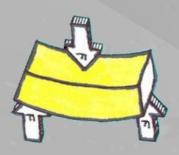
Stress σ:

according the type of loading



compression tension

$$\sigma = \frac{F}{A}$$



bending

$$\sigma = \frac{M}{W}$$



Young's modulus determination

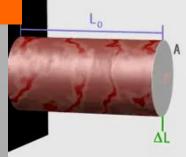
• strain ε

$$\varepsilon = \frac{\Delta I}{I_0} = \frac{I_1 - I_0}{I_0}$$

ΔI change of the length [mm]

I₁ length after elongation [mm]

I₀ original (initial) length [mm]







Measuring of elongation ΔI

 deformations ∆I have to be measured by special devices - strain gauge

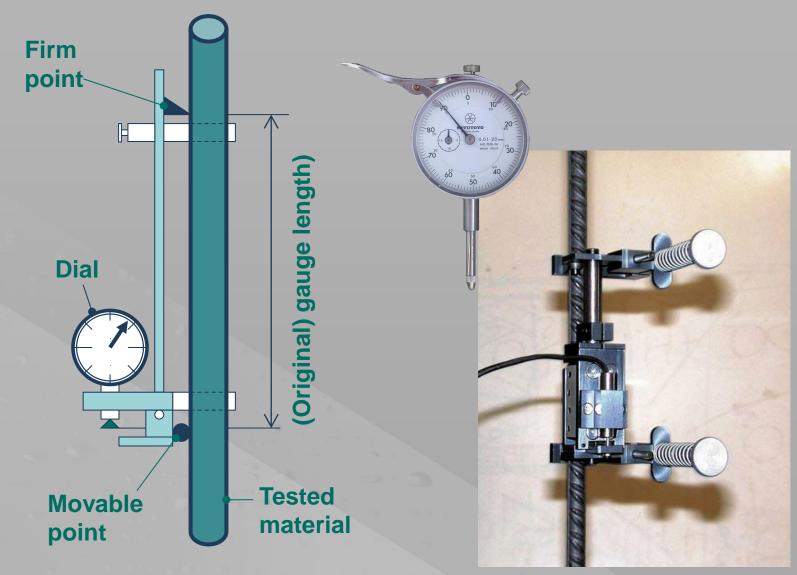
Strain gauge:

- -mechanical
- -electrical
- -optical



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Mechanical strain gauge

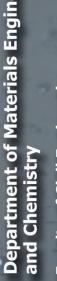


Measuring of deformations

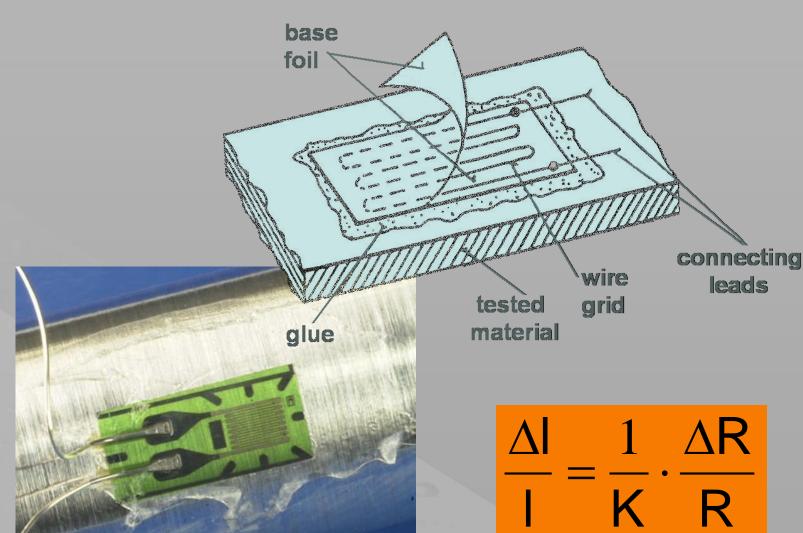




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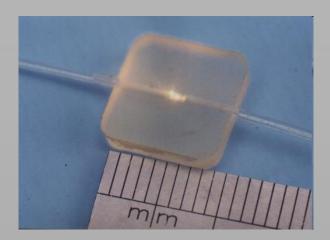


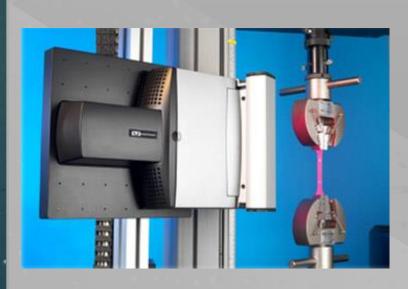
Electrical resistance gauge

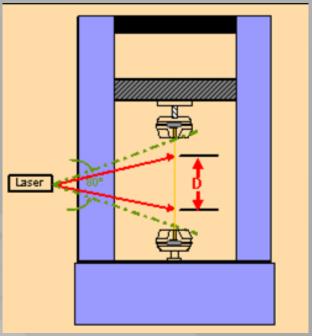


Optical strain gauge

- optical fibers
- laser









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Dynamic Young's modulus

ultrasonic waves

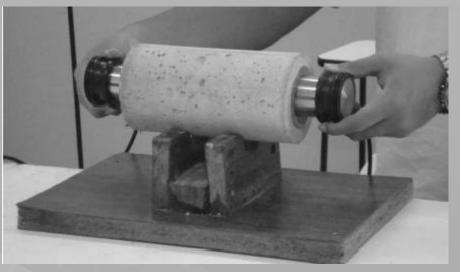
$$E_{dyn} = c^2 . \rho_v$$

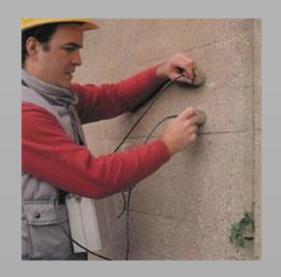


sound velocity [m².s⁻¹]

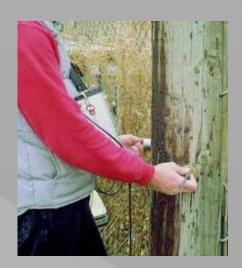
bulk density [kg.m⁻³]

Measuring of dynamic Young's modulus











Elastic modulus is affected by:

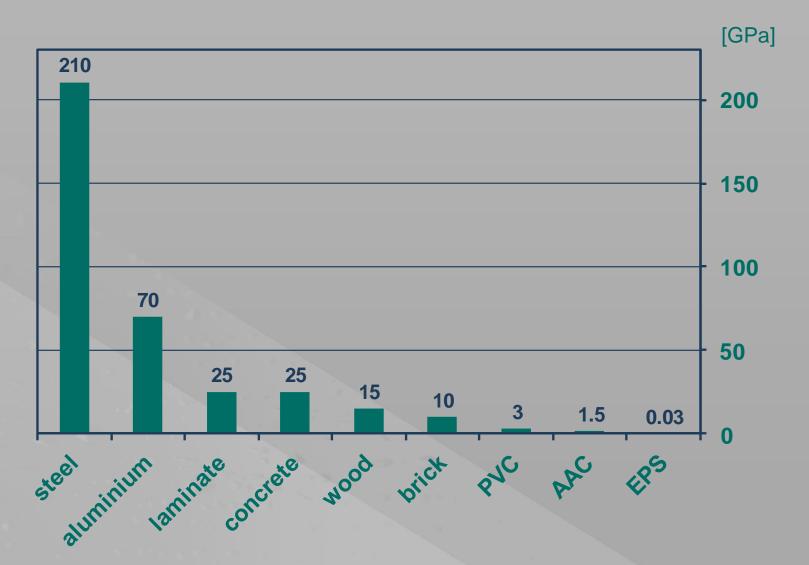
- temperature
 - thermoplastics: with rising temperature
 E significantly decreases
 - concrete:

-20 °C - +70 °C - E constant under -50 °C - ca. about 20 % higher above 300 °C - ca. 50 % of initial value

moisture – in porous materials

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Young's modulus of some materials



Material	Young's modulus [GPa]
Diamond	1050-1200
Steel	210
Glass	50 -85
Aluminium and light alloys	65 -73
Brass and bronze	103-124
Concrete	15 - 60
Ceramic brick	8 - 12
Wood	7 -18
Glass laminate	10 - 30
Thermosets	4 - 13
Thermoplastics solid	0,1 - 4
Thermoplastics foamed	0,02-0,3
Rubber	0,002 -0,005

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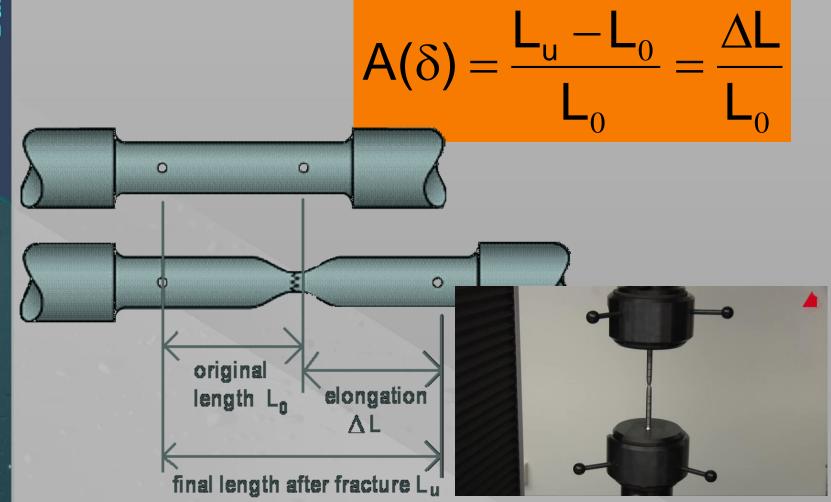
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Building materials

Ductility

percentage elongation after tensile test



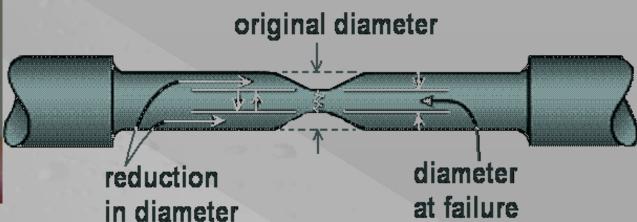
Reduction of area

change of cross sectional area as a percentage of the original cross-sectional area

$$Z(\psi) = \frac{S_0 - S_u}{S_0} \quad \begin{array}{l} S_0.... & \text{original cross-sectional} \\ \text{area before testing} \\ S_u.... & \text{minimal cross-sectional} \end{array}$$

area after failure





Brittle x tough materials

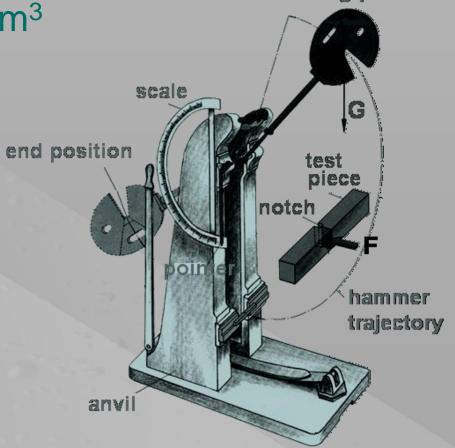
- brittle material, subjected to stress, breaks without significant deformation
- tough material deforms plastically and absorbs energy before fracture



Toughness

 the amount of energy per volume that a material can absorb before rupturing starting position

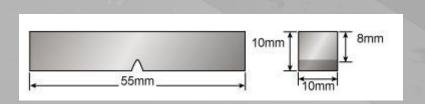
units: kJ/m³



Test of toughness

- impact toughness
 - Charpy, Izod test
- notch toughness

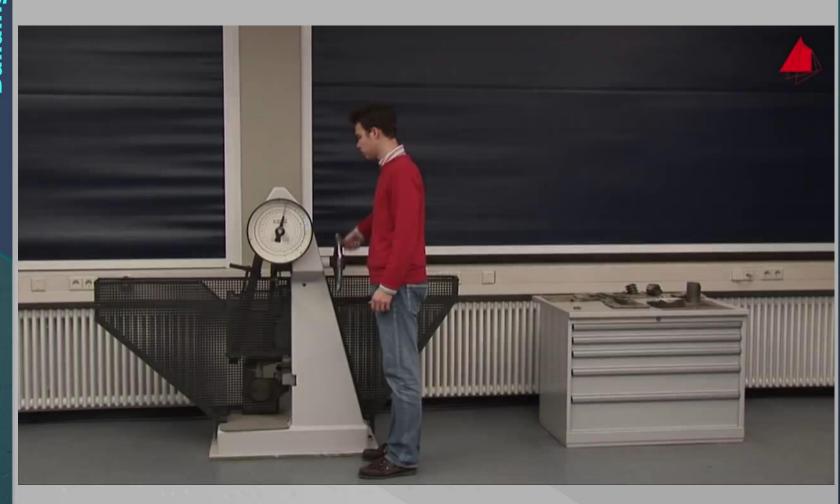
 (ability to absorb energy in the presence of a flaw)





dina material

Charpy impact test



Brittleness

- tendency of a material to fracture or fail upon the application of a relatively small amount of force, impact, or shock
- opposite of toughness
- no numerical value



Rough criterion for brittle materials:

compressive strength: tensile strength

> 8:1

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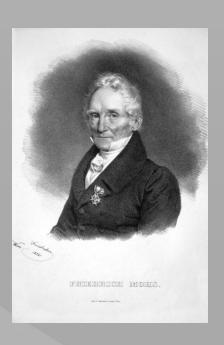


Building materials

Hardness

- defines the materials resistance to penetration
- depends on temperature and moisture

- Methods:
 - scratch hardness
 - indentation hardness
 - rebound hardness



Scratch h. - Mohs scale

- 1. talc
- 2. gypsum
- 3. calcite
- 4. fluorite
- 5. apatite
- 6. feldspar (orthoclase)
- 7. quartz
- 8. topaz
- 9. corundum
- 10. diamond
 - used for minerals







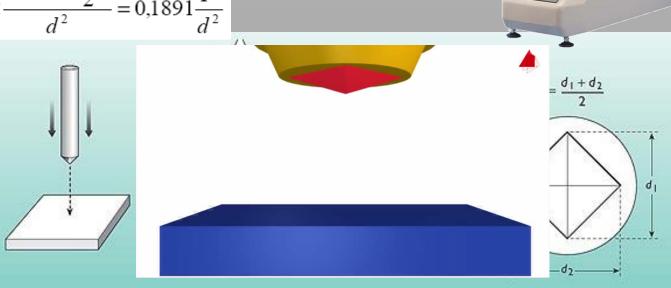




Indentation h. - Vickers test

- indenter: diamond point with a 136° point angle
- abbreviation VHN
- metals, hard materials

$$HV = 0.102 \times \frac{2F \sin \frac{136^{\circ}}{2}}{d^2} = 0.1891 \frac{F}{d^2}$$

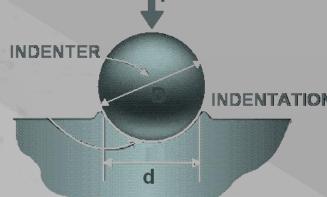


Indentation h. - Brinell test

- indenter: steel (tungsten) ball (10 mm ∅)
- abbreviation: HBW, (HBS)
- metals, wood, hard polymers

 $HB = 0.102 \frac{2F}{\pi . D \cdot (D - \sqrt{D^2 - d^2})}$

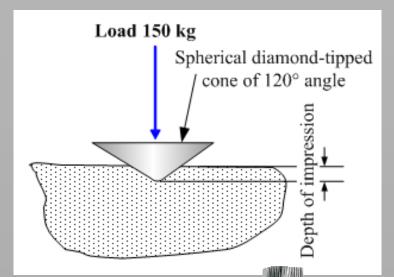






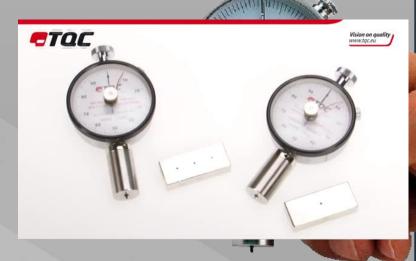
Indentation h. – Rockwell test

- diamond cone
- abb.: **HR**(A,B,C..G)
- depth of indentation
- metals



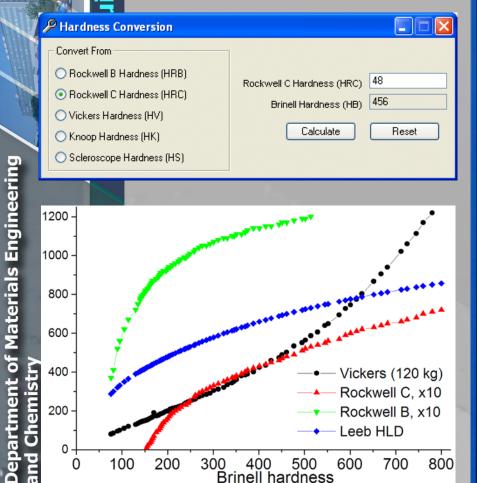
Shore durometer

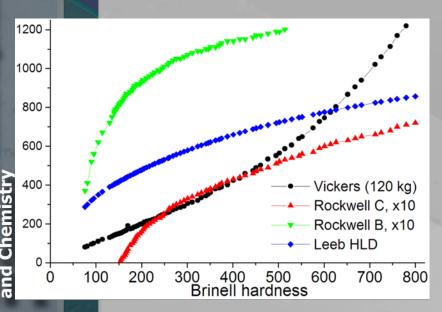
- spring+ steel rod
- abb.: **SH**
- polymers, elastomers, rubber



Hardness conversion

indicative only





HARDENED STEEL AND HARD ALLOYS

Rockwell*				Superficial			Vickers	Knoop	Brinell	Tensile Strength	Micro- ficial
C	Α	D	G	15-N	30-N	45-N	HV	HK	НВ	KSI	WMN
150 kg Brale	60 kg Brale	100 kg Brale	150 kg 1/16" ball	15 kg N Brale	30 kg N Brale	45 kg N Brale	10 kg	500 gm and over	3000 kg 10 mm ball	1000 lbs/ sq in	1000 gm
80 79 78	92.0 91.5 91.0	86.5 85.5 84.5		96.5 96.3 96.0	92.0 91.5 91.0	87.0 86.5 85.5	1865 1787 1710	111	•	A	111
77 76 75	90.5 90.0 89.5	84.0 83.0 82.5		95.8 95.5 95.3	90.5 90.0 89.0	84.5 83.5 82.5	1633 1556 1478				
74 73 72	89.0 88.5 88.0	81.5 81.0 80.0		95.0 94.8 94.5	88.5 88.0 87.0	81.5 80.5 79.5	1400 1323 1245	111	NOTE 1	NOTE 2	Ξ
71 70 69	87.0 86.5 86.0	79.5 78.5 78.0		94.3 94.0 93.5	86.5 86.0 85.0	78.5 77.5 76.5	1160 1076 1004	972 946			953 949
68 67 66	85.6 85.0 84.5	76.9 76.1 75.4		93.2 92.9 92.5	84.4 83.6 82.8	75.4 74.2 73.3	940 900 865	920 895 870	NA NA		945 942 938
65 64 63	83.9 83.4 82.8	74.5 73.8 73.0		92.2 91.8 91.4	81.9 81.1 80.1	72.0 71.0 69.9	832 800 772	846 822 799	739 722 706		934 930 926
62 61 60	82.3 81.8 81.2	72.2 71.5 70.7		91.1 90.7 90.2	79.3 78.4 77.5	68.8 67.7 66.6	746 720 697	776 754 732	688 670 654	 NA	922 917 913
59 58 57	80.7 80.1 79.6	69.9 69.2 68.5		89.8 89.3 88.9	76.6 75.7 74.8	65.5 64.3 63.2	674 653 633	710 690 670	634 615 595	351 338 325	909 904 900
56 55 54	79.0 78.5 78.0	67.7 66.9 66.1		88.3 87.9 87.4	73.9 73.0 72.0	62.0 60.9 59.8	613 595 577	650 630 612	577 560 543	313 301 292	896 891 887
53 52 51	77.4 76.8 76.3	65.4 64.6 63.8		86.9 86.4 85.9	71.2 70.2 69.4	58.6 57.4 56.1	560 544 528	594 576 558	525 512 496	283 273 264	883 879 874
50 49 48	75.9 75.2 74.7	63.1 62.1 61.4		85.5 85.0 84.5	68.5 67.6 66.7	55.0 53.8 52.5	513 498 484	542 526 510	481 469 455	255 246 238	870 865 861
47	74.1	60.8		83.9	65.8	51.4	471	495	443	229	856

Indentation h.—POLDI hammer

Comparison of the indentation size of tested material and reference material with known hardness



Rebound h. – Schmidt hammer

 measures the rebound of a springloaded mass impacting against the surface of the sample





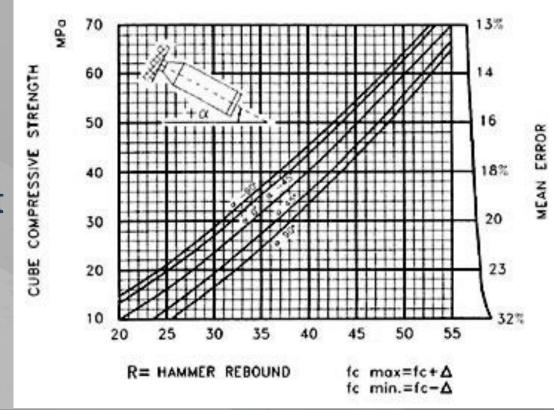


Correlation between Schmidt rebound number and the compressive strength

 the rebound value can be used to determine the compressive strength (by reference to the conversion chart)

Depends on:

- orientation of the hammer
- water content



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Building materials



Fatigue



- fatigue occurs when a material is subjected to repeated loading and unloading
- cyclic stress causes the decrease of the strength
- typical for metals

Fatigue limit (strength) = the amplitude (or range) of cyclic stress that can be applied to the material without causing fatigue failure

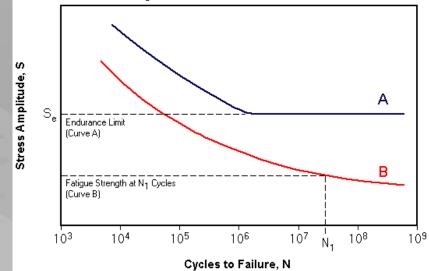
Fatigue

- if the loads are above a certain threshold, microscopic cracks will begin to form
- after reaching critical size, and the structure will suddenly (without warning) fracture
- the shape of the structure affect the fatigue life (square holes, sharp corners)
- the greater the applied stress range, the shorter the life
- damage is cumulative, materials do not recover when rested
- f. is influenced by a variety of factors (temperature, surface finish, microstructure, presence of oxidizing or inert chemicals, residual stresses, etc.)

Endurance limit

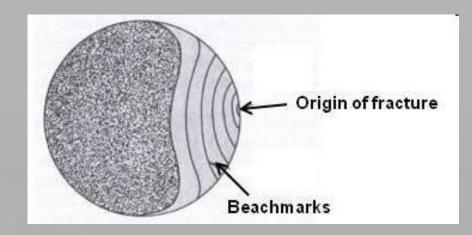
- some materials (ferrous and titanium alloys)
 have a distinct limit below which there
 appears to be no number of cycles that will
 cause failure
- some structural metals (aluminium, copper)
 do not have a distinct limit and will eventually
 fail even from small stress amplitudes

S-N (Wöhler) curves

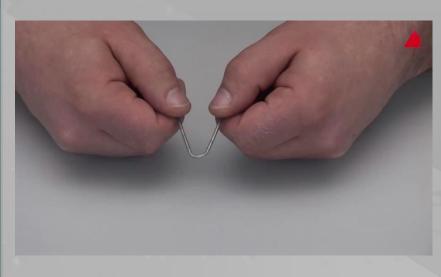


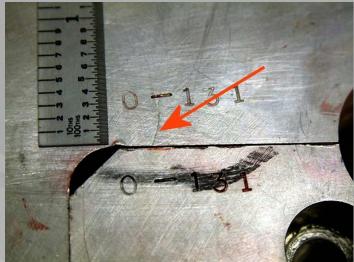
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Fatigue cracks



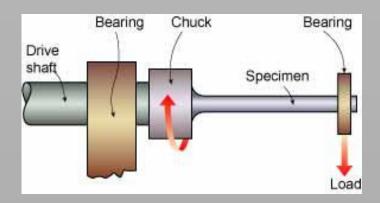


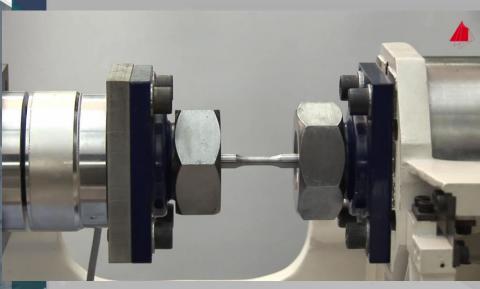


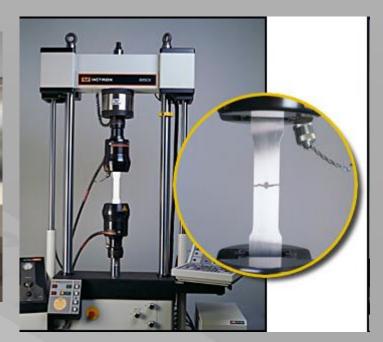


Fatigue testing









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Infamous fatigue failures

Boston Molasses Disaster (USA, Boston, 1919)





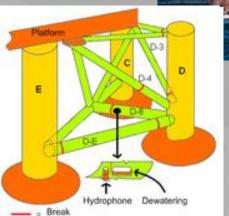
Infamous fatigue failures

 Alexander L. Kielland oil platform capsize (Norway, 1980)







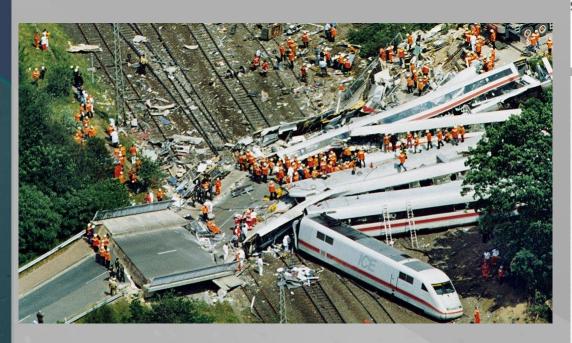


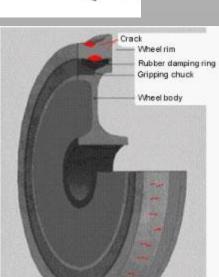


Infamous fatigue failures

• InterCity expres (Germany, Eschede, 1998)

fatigue crack in one wheel





Dynamic strength

 Tacoma narrows bridge (USA, Washington, 1940)



Abrasion resistance

- ability of a surface to resist being worn away by rubbing or friction
- coatings, paints, floor surfaces, pipes

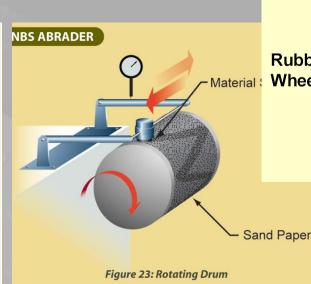


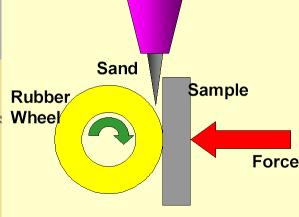
Abrasion resistance tests

 usually measured as a loss percentage based on original weight

big scale of different tests



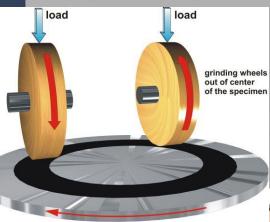




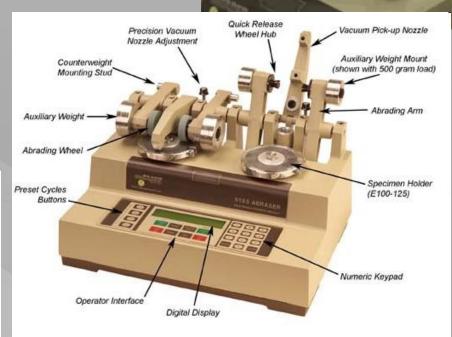
Materials Engineering

Abrasion resistance tests Taber abrader

 thickness loss after defined number of rotations using standardized wheel and defined load







Comparative Abrasion Resistance of Various Polymers

Material	Weight Loss (mg)			
thermoplastic urethane	0.4-3.2			
ionomer	12			
nylon 6/10	16			
nylon 11	24			
HDPE	29			
polytetrafluorethylene	42			
nitrile rubber	44			
nylon 6,6	58			
LDPE	70			
rigid PVC	122			
natural rubber (tread formulation)	146			
SBR (premium tread formulation)	177			
SBR (tread formulation)	181			
plasticized PVC	187			
butyl rubber	205			
ABS	275			
neoprene (polychloroprene)	280			
polystyrene	324			
Taber abrasion, CS17 wheel, 1000 gm weight, 5000 revolutions Ref: Handbook of Thermoplastic Elastomers, Litton Educational Publishing,				



Abrasion resistance tests Amsler/Böhme test

- a specimen is subjected to stress by grinding
- the abrasive grit accumulating from this is indicated as loss of volume or thickness (abrasive loss) per test area 50 cm²





test area 7,07 x 7,07 cm



Abrasion resistance - standards (Amsler / Böhme)

- EN 1338, 1339, 1340 concrete pavings
- EN 14157 natural stones
- EN 13 892-3 screeds

abrasion resistance cm³/50 cm²						
1	00	80	60	40	20	
Granite + Tonalite						w
Syenite + Gabbro					52	~
Lava						
Vulcanic tuff			2000	***	****	
Sandstone					*****	
Clay shale	****	*****	******	******	***	
Limestone compact				5000	888	
Limestone porous	****	*****	******	2000)		
Marble				500	*****	
Serpentinite						8
Paragneiss					500	×
Quarzite					5000	1

Fig. 3.70 Böhme test abrasion values for different rock types including their range of dispersion (modified after Müller 2001). Note, higher values (from right to left) indicate a rock having a lower abrasion resistance, lower values indicate a higher abrasion resistance

Underwater Method

 ASTM C1138: Standard Test Method for Abrasion Resistance of Concrete







Set-up sample for apparatus



Test in progress - Between Cycles





Sample removed from Apparatus - Between Cycles



Sample Between Cycles



Sample After 6 Cycles (Each Cycle is 12 Hours)



Adhesion and cohesion

- adhesion state in which two surfaces are held together by interfacial effects
- cohesion ability of a material to maintain its strength when unconfined

adhesion < cohesion

Adhesion

 between two materials - in composites (steel + concrete, cement + aggregates...)







Adhesive strength

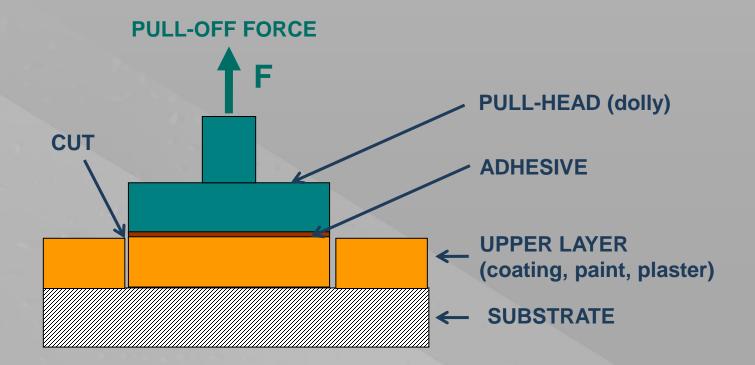
 between the upper layer and base (plasters, coatings..)





Pull-off test

 the circular pull-head plates are glued to the test material and upper layer is cut around them



Pull-off test equipment







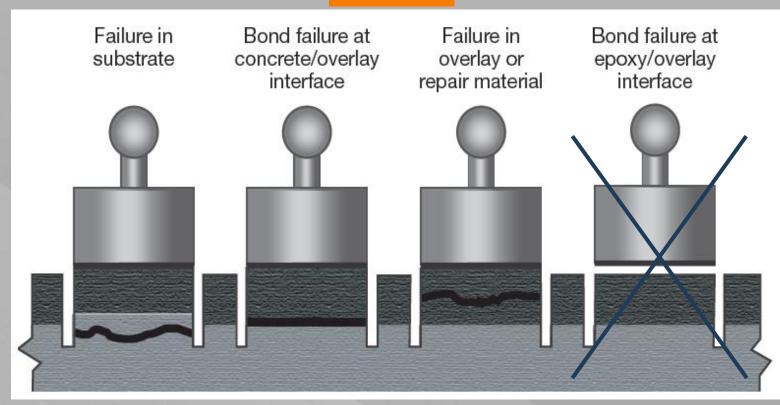
Pull-off Adhesion Testing Instructional Video

Adhesive strength fa

test results:

 $f_a > f_u$

$$f_u = \frac{F_u}{A}$$



invalid result

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Building materials

Moisture (hygric) properties







Definitions

- Moisture content amount of water contained in a material
- Dampness presence of unwanted moisture in the structure of a building,
- Humidity amount of water vapor in the air
- Hygric relating to moisture



Moisture content

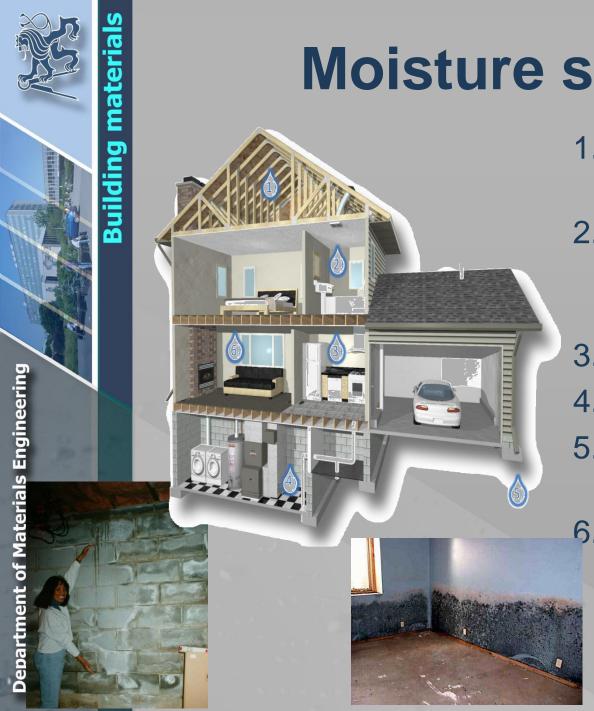
 amount of water contained in a material, which can be removed by

drying

 any porous material in the construction is not quite dry!

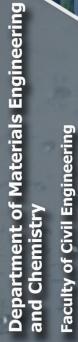


: AND THE PROPERTY LIES IN A BEAUTIFUL VALLEY BY THE BANKS OF THE RIVER ...'



Moisture sources

- 1. Leaking roofs and ice dams
- 2. High humidity in bathrooms and kitchens
- 3. Leaking pipes
- 4. Flooding in basement
- 5. Pooling water at foundation
- 6. Condensation on windows and exterior walls

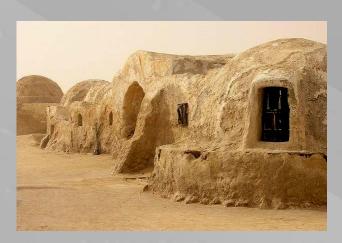


	Source	Moisture Production Liters per Day
	People (Evaporation per Person)	0.75 (Sedate), 1.2 (Avg.) to 5 (Heavy Work)
	Humidifier	2–20+
	Hot Tub, Whirlpool	2–20+
	Firewood, per Cord	1–3
	Washing Floors etc.	0.2
	Dishwashing	0.5
	Cooking for Four	0.9 to 2 (3 with Gas Range)
	Frost-Free Fridge	0.5
	Typical Bathing/Washing per Person	0.2 to 0.4
	Shower (ea.)	0.5
	Bath (ea.)	0.1+
	Unvented Gas Appliance	0.15 kg/kWh for Natural Gas, 0.10 kg/kWh for Kerosene
	Seasonal Desorption (or New Materials)	3 – 8 Depends on the House Construction
	Plants/Pets	0.2-0.5 (Five Plants or One Dog)

Moisture types

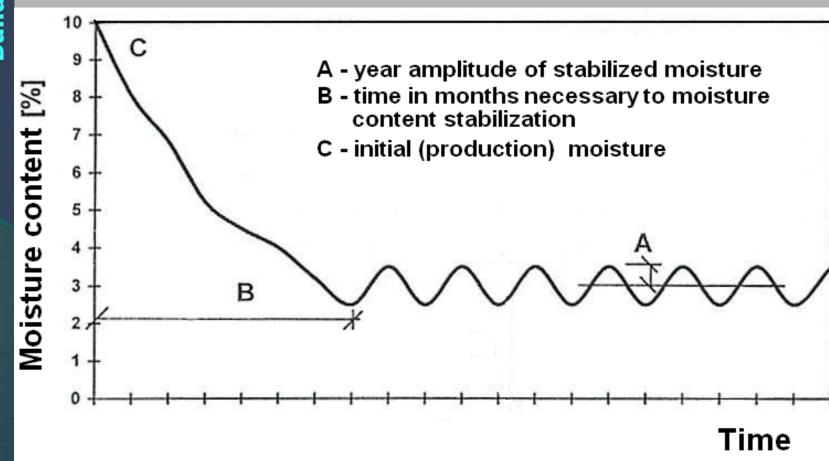
According time:

- initial moisture (natural, manufacturing)
- storage moisture
- stabilized moisture (constant after longer period - ca 2 - 7 years)





Moisture content during time period





Gravimetric water content

(Water content by mass)

$$\mathbf{w}_{m} = \frac{\mathbf{m}_{H_{2}O}}{\mathbf{m}_{D}} = \frac{\mathbf{m}_{W} - \mathbf{m}_{D}}{\mathbf{m}_{D}} (*100)$$

mass of dry material \mathbf{m}_{D}

mass of wet material

mass of water in material **m**_{H₂O}

Volumetric water content

(Water content by volume)

$$\mathbf{w}_{V} = \frac{\mathbf{V}_{H_{2}O}}{\mathbf{V}} = \frac{\mathbf{m}_{W} - \mathbf{m}_{D}}{\rho_{H_{2}O} * \mathbf{V}} (*100)$$

mass of dry material

mass of wet material

volume of water in material **V**_{H₂O} ...

volume of material

Maximal water content

- amount of water taken in by a material when immersed
 - material is saturated (fully soaked)
- volumetric water content
 - **0 100 %**
- gravimetric water content:
 - could be higher than 100 %
 (lightweight materials)



Maximal water content x porosity

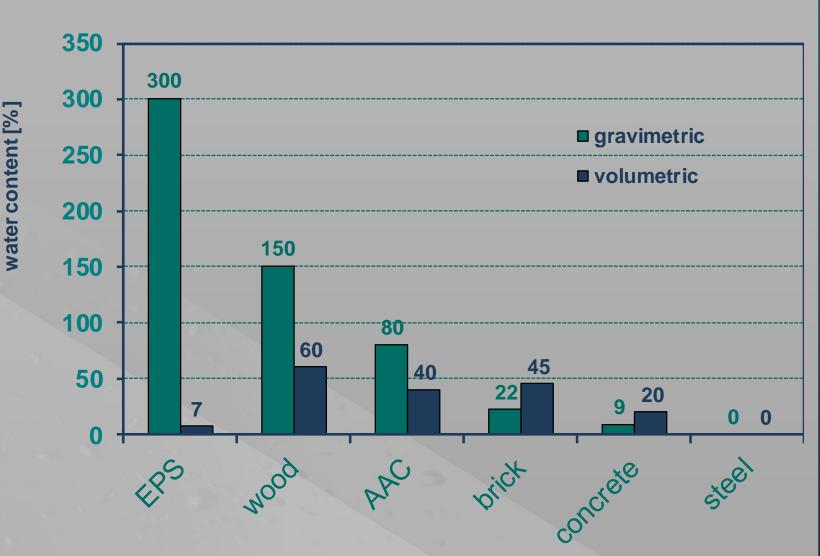


- when material is saturated, water fills up all
 open pores →
 - value of maximal volumetric water content
 = value of open porosity

 $\mathbf{w}_{\text{max}} = \mathbf{p}_{\text{open}}$

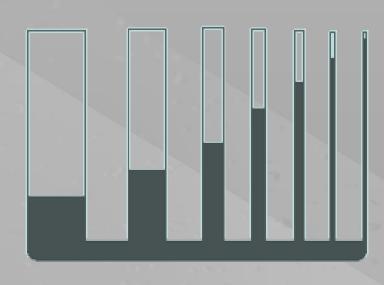
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Maximal water content



Capillarity

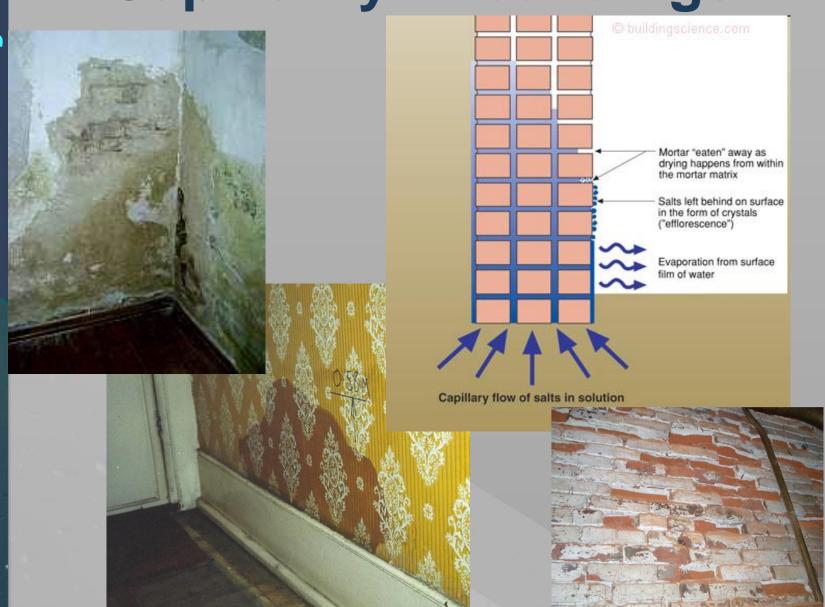
 ability of a liquid to flow in narrow spaces without the assistance of, and in opposition to external forces like gravity





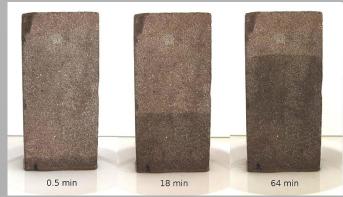
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Capillarity in buildings



Capillary water absorption test

 the material is put into the low level of water (its sides are sealed) and increase of its mass is measured at given times



Sealed with paraffin wax Water Inhibitor or			0.5 min 18 min 64 min			
			Bulk density in dry state	Gravimetric capillarity [kg/m³] after		
ر ر		no treatment	[kg/m³]	2 hours	8 hours	24 hours
ot Materia ry Engineering		Foamed polystyrene	16	1,1	1,2	1,2
tment nemist		Foamed polyurethane	35	0,3	0,4	0,4
Depari and Ch Faculty		AAC	540	12,7	19,9	29,7

Sorption behavior

 ability of a hygroscopic material to absorb or release water vapor from or into the air until a state of equilibrium is reached

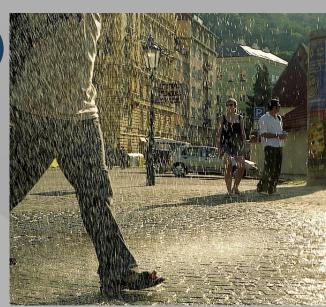
adsorption

(uptake of water vapor)

X

desorption

(water vapor release)

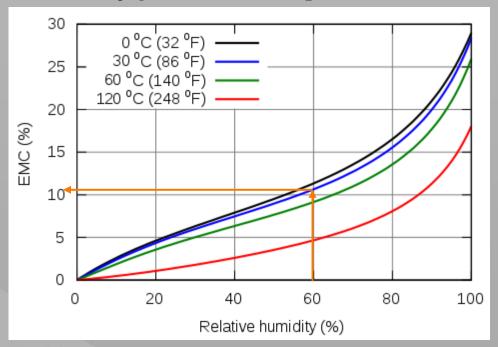




Equilibrium moisture content

each material has typical sorption

isotherms

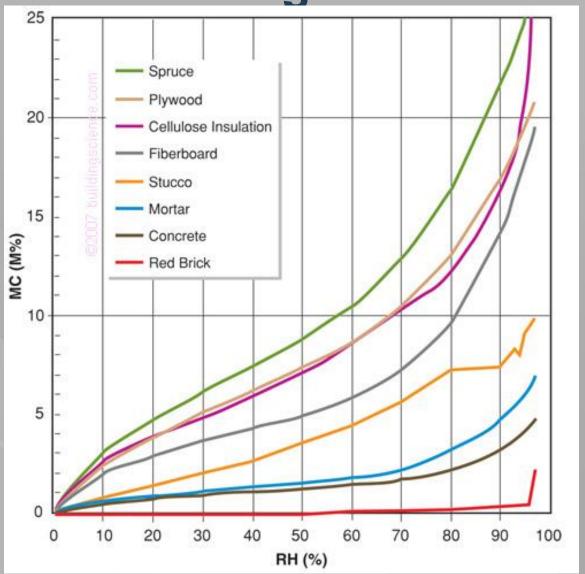


Depends on:

- · air temperature, pressure and humidity
- pores sizes and shapes
- history

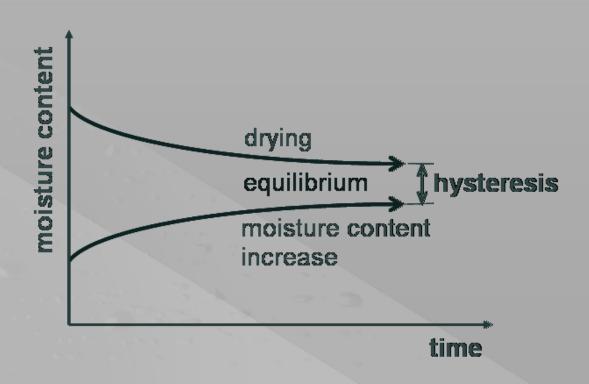
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Sorption isotherms of some building materials



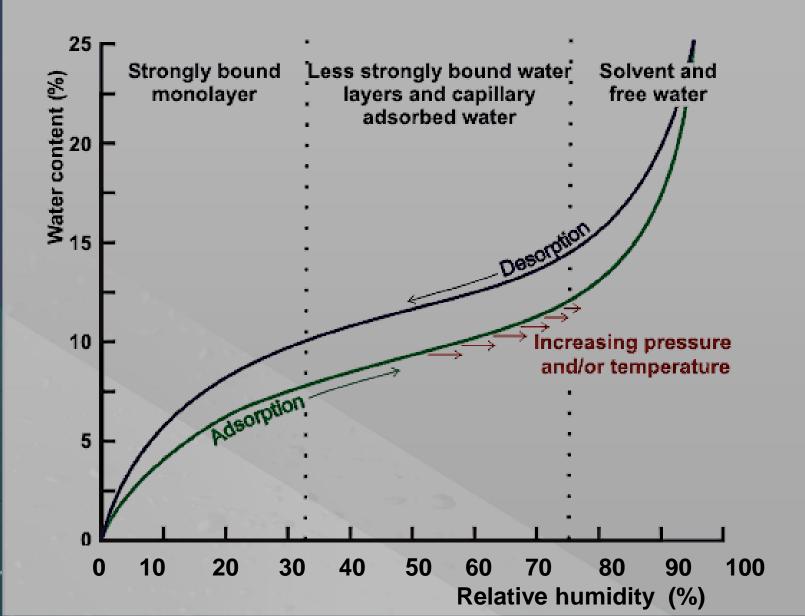
Sorption hysteresis

 desorption give higher equilibrium moisture contents than absorption at equal ambient climate conditions



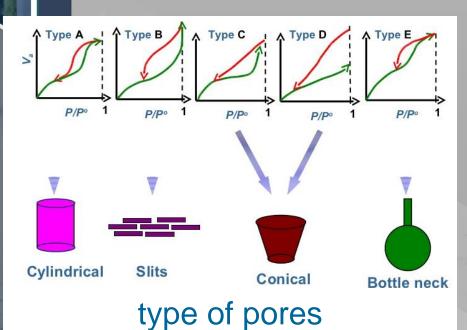
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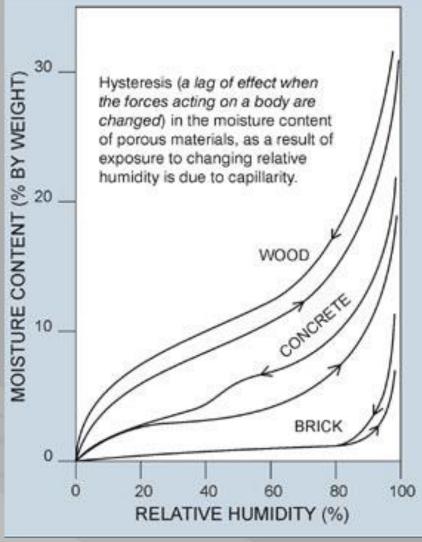
Sorption hysteresis curve



Sorption hysteresis

 shape of hysteresis curve depends on the shape of pores





Water permeability





Water permeability

- roofing (ceramic and concrete tiles))
- depends on the amount of capillary pores and cracks (size 0,01 – 0,5 mm)

Roof tiles are always permeable

space under roof has to be ventilated!

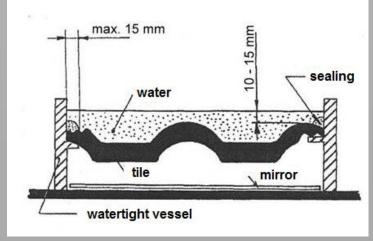


Water impermeability test

measuring the time taken for a drop of water to appear on the underside of the roofing tile when a 6 cm thick water head is applied on

the opposite side





measuring the volume of water that passes through a saturated roofing tile under a 10 cm thick water head

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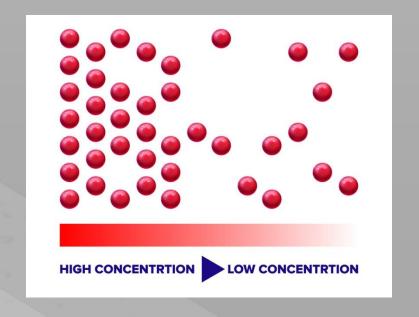
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Building materials

Diffusion

- gases, vapor
- spread of particles through random motion from regions of higher concentration to regions of lower concentration
- Important for:
 - vapor barriers
 - radon barriers
 - rehabilitation renders
 - paints
 - passive houses





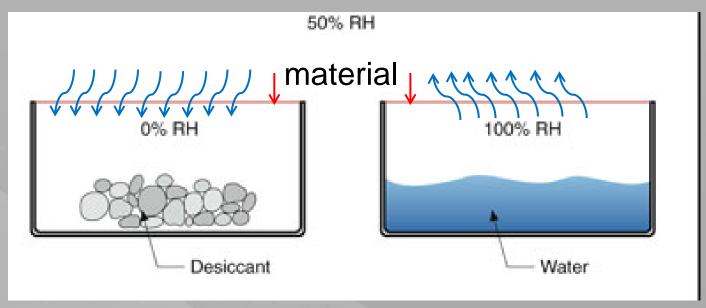
Water vapor transport

Important properties:

- the water vapor diffusion coefficient δ
- the resistance to water vapor diffusion factor µ
- the water vapor diffusion equivalent air layer thickness S_d

Water vapor permeability test

- dry cup method wet cup method



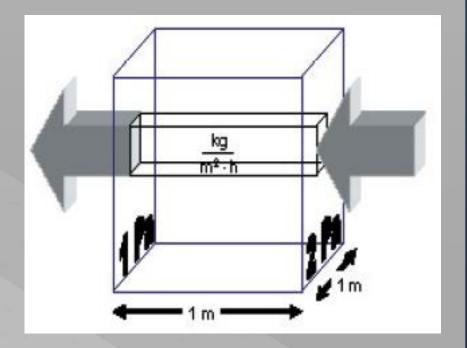
mass increase

mass decrease

Water vapor diffusion - coefficient δ

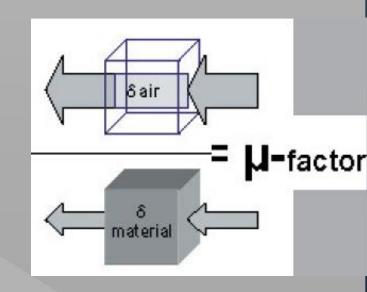
 the amount of water vapor [kg] which diffuses through a layer of material which is 1 m thick and has an area of 1 m² at a partial water vapor pressure difference of 1 Pa in 1 hour

units: [kg/m.h.Pa]



Resistance to water vapor diffusion - factor μ

- the ratio of the water vapor diffusion coefficient of the air δ_L to the value δ_{mat} of the material
 - a measure for the vapor tightness of a material (how many times greater the resistance to transmission of a layer of the material is compared to a static layer of air of the same thickness)



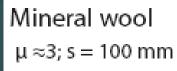
μ -value of some materials

Material	μ -value	
Reinforced concrete	90	
AAC (autoclaved aerated concrete)	6	
Masonry of full bricks	7	
Steel (plate)	600 000	
Window glass	10 000	
Gypsum board	12	
Gypsum fibre board	10 -15	
Wooden fiber plates	10	
Hydrophobic plywood	60 - 100	
Mineral wool	2	
OSB (oriented strand board)	30	
EPS (expanded polystyrene)	40	
XPS (extruded polystyrene foam)	170 - 200	

Equivalent air layer thickness S_d

$$S_d = \mu \cdot s [m]$$

the thickness of a static layer of air [m], which displays the same resistance to water vapor transmission as the building material in the thickness s with the resistance to water vapor transmission value µ



Polyurethane $\mu \approx 100$; s = 100 mm

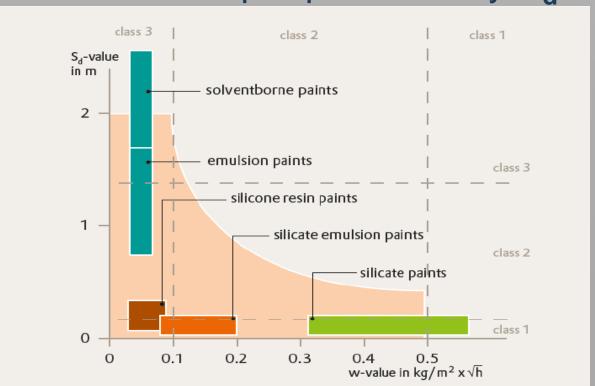
$$S_{d} = 0.3 \text{ m}$$

- Sd ≤ 0.5 m => **diffusion**open materials
- $0.5 \text{ m} < \text{Sd} \Rightarrow \text{diffusion}$ blocking materials
- Sd ≥ 1500 m ≥ 1500 m => diffusion-proof materials

Diffusion of coatings

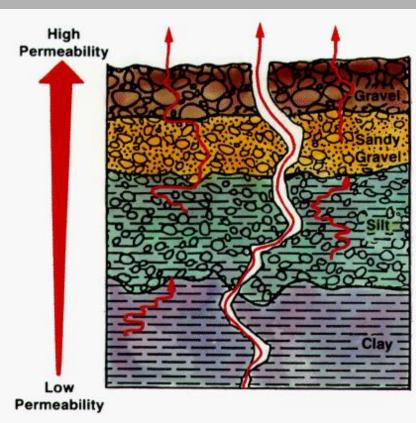
Künzel's facade protection theory:

 protection and breathability of coating are influenced by water absorption capacity w and the water vapor permeability S_d



Soil permeability

- ability of soil to transmit water and air
- important in the areas with radon risk





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Building materials

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uilding materials

Thermal properties



Interaction heat - material

If the surroundings of the material has different temperature than the material, the thermal energy is transferred

Types of interaction:

- 1. The properties of material influence the transmission of heat
- 2. Thermal energy influences the properties of material



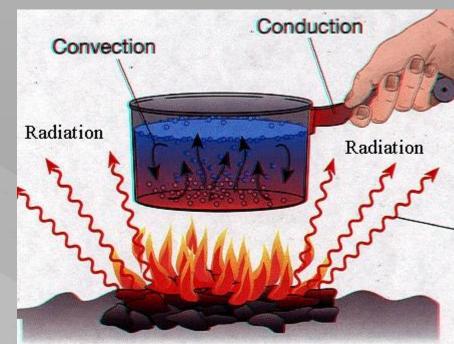
Heat transport

- conduction solids, gases, liquids
 - the transfer of heat within a substance, molecule by molecule
- convection gases, liquids

- heat transfer by the mass movement of a fluid in

the vertical direction

- radiation gases
 - heat is transfered through wave energy

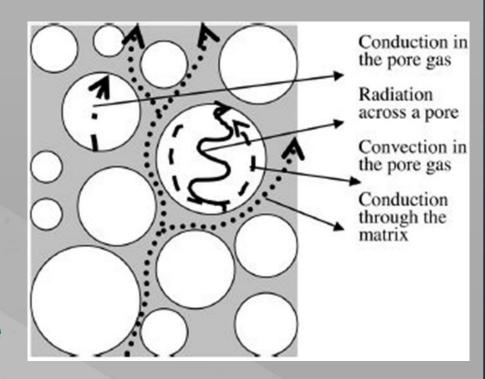


Heat transport

 In the porous materials the heat is transferred by combination of all types of heat transport

Depends on:

- porosity
- structure
- temperature
- material type



Thermal conductivity

a material's ability to conduct heat



(coefficient of) thermal conductivity

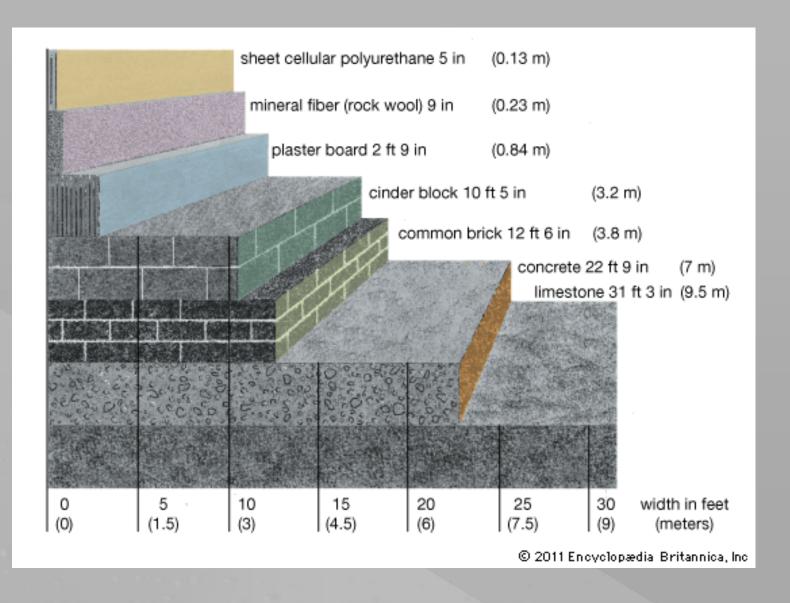
Thermal conductivity

- the quantity of heat transmitted, due to unit temperature gradient, in unit time under steady conditions in a direction normal to a surface
- λ (K-value)
- units SI: [W/m.K]
 - imperial units: [Btu/hr.ft.F] (1 Btu/hr.ft.F = 1.730735 W/m.K)
- the lower λ, the better insulator
 - (thermal insulating materials λ < 0,15 W/m.K)
- range λ : 10⁻² 10² W/m.K



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Thermal conductivity - comparison



Thermal conductivity **Basic facts**

- Organic materials are better insulators than inorganic m.
- Crystalline materials are better conductors than amorphous m.
- Materials with lower bulk density are usually better insulators.
- Metals are very good conductors.
- Anisotropic materials have different conductivity in different directions.



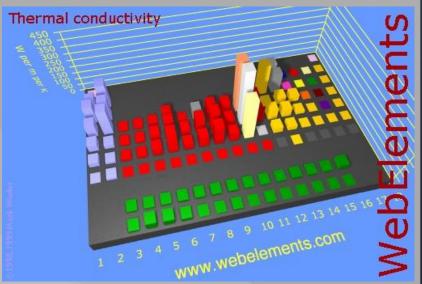




Thermal conductivity

Depends on:

- chemical composition
- structure
- porosity (bulk density)
- moisture
- temperature



materials

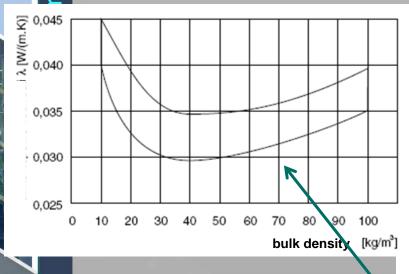
Influence of porosity on thermal conductivity

- $\lambda_{AIR} = 0.025 \text{ W.m}^{-1}.\text{K}^{-1}$
- the higher the amount of air in the material (porosity), the lower bulk density and thermal conductivity is
- size of pores is limited (best 0,1 1 mm)
 (because of capillarity)

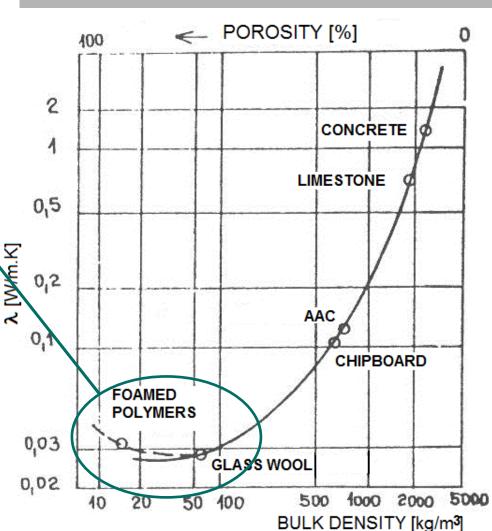


aterials

Porosity x thermal conductivity







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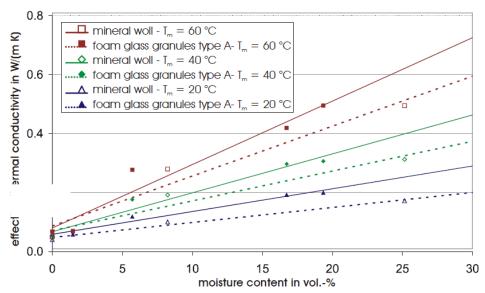
Influence of moisture on thermal conductivity

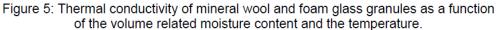
 λ_{water} app. 25 x higher than λ_{air}

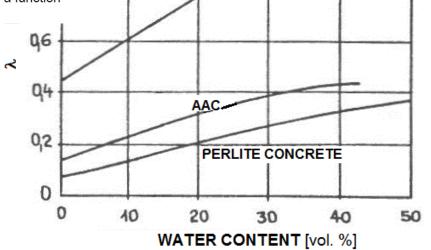




- moisture significantly reduces the thermal insulating ability of materials
- very small pores are liable to wetting (capillarity) - the best size of pores in insulating materials: 0,1 - 1 mm



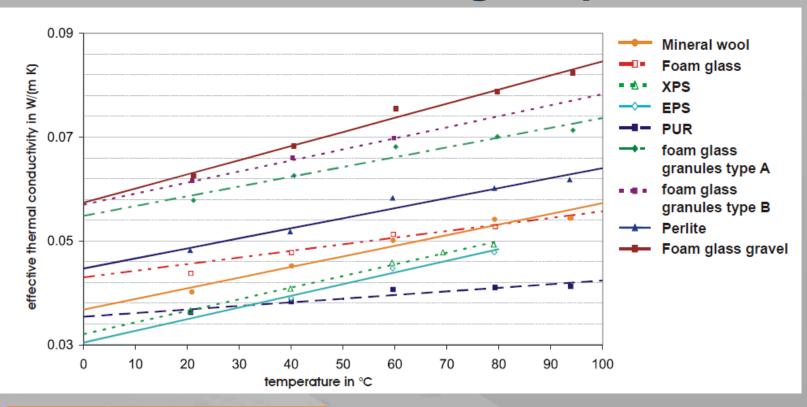




BRICK

Influence of temperature on thermal conductivity

λ increases with rising temperature



$$\lambda_t = \lambda_0 + 0.0025 t$$
 (for t = 0 - 100°C)



Thermal insulating materials overview

- fibrous mineral and glass fibers



- wood wool (excelsior)
- cellulose fibers
- recycled paper fibers
- straw (bales, loose)

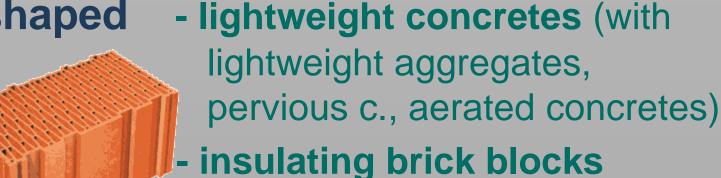
porous particles



- expanded clay aggregate
- expanded perlite
- ash
- cinders

Thermal insulating materials

shaped



- diatomite

foamed



- foamed polymers (PU,PS, phenolic foam)
- cellular glass

Thermal insulating materials

other

- cork (expanded)
- wood
- wood-based materials
 (fiber board, particle board)

- lamb wool



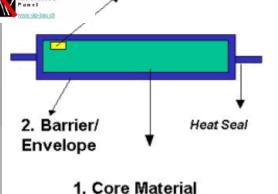


Best insulation?



- vacuum
- "VIP" = Vacuum Insulated Panel
- a nearly gas-tight enclosure surrounding a rigid core, from which the air has been evacuated
- $\lambda = 0.004 0.008 \text{ W/m} \cdot \text{K}$





inside Getter/Desiccant

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Common available insulation materials

INSULATION MATERIAL

THERMAL CONDUCTIVITY W/MK - LESS INDICATES **BETTER PERFORMANCE**

THERMAL RESISTANCE M2K/W - MORE **INDICATE BETTER PERFORMANCE**

U - VALUE W/M2K

Mineral Wool			
Glass fibre	0.032 - 0.044	3.10 - 2.25	0.32 - 0.44
Rock fibre	0.035 - 0.044	2.85 - 2.25	0.35 - 0.44
Sheep's wool	0.042	2.38	0.42
Expanded polystyrene (EPS)	0.036	2.77	0.36
Multi-foils	0.040	The nature of this insulant does not lend its directly to direct comparison on thermal resistance or stand alone U-value	
Hemp	0.039	2.56	0.39
Extruded polystyrene (XPS)	0.029 - 0.036	3.44 - 2.77	0.29 - 0.36
Polyurethane foam board (PUR)	0.22 - 0.29	0.45 - 3.44	2.22 - 0.29
Polyisocyanurate foam board (PIR)	0.021 - 0.022	4.76 - 4.54	0.21 - 0.22
Phenolic foam board	0.021	4.76	0.21
Evacuated panels - 20mm thk	0.004	5.00	0.20
Aerogel board - 10mm thk	0.013	0.77	1.29

Thermal resistance figures are based 100mm thickness of insulation material Thermal conductivity figures are typical for each material and may vary slightly between manufacturers U - values shown indicate heat loss in watts / sq.m degree for insulation product alone

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Thermal conductivity measuring

- steady-state methods the temperature of the measured material does not change with time
- transient (non steady-state) methods a measurement during the process of heating up

Steady state - a situation in which all variables are constant in spite of ongoing processes that strive to change them. For an entire system to be at steady state, i.e. for all state variables of a system to be constant, there must be a flow through the system

Thermal conductivity measuring

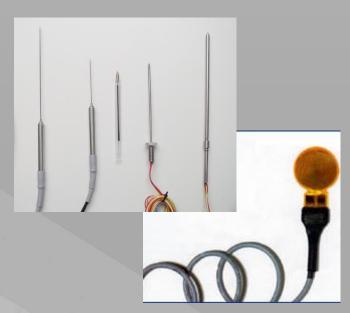
Steady-state methods:

- Guarded hot plate
- Divided bar
- Hot box

Transient methods:

- Hot wire
- Plane source
- Needle probe
- Laser flash method





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Thermal conductivity measuring











Thermal conductivity determination

Steady state method:

$$\lambda = \frac{q \times d}{T_1 - T_2}$$

$$q = \frac{Q}{A}$$

of heat passing through a unit area of the sample in unit time [W/m²]

d average thickness of sample [m]

T₁ temperature of warm side of the sample [K]

T₂... temperature of cold side of the sample [K]

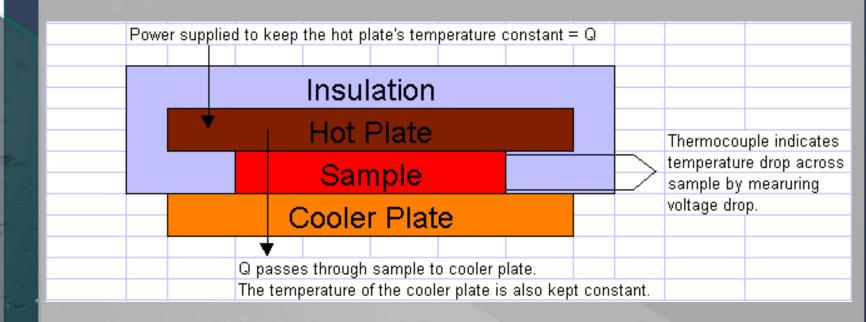
Q quantity of heat passing through a base area of

the sample [W]

A base area of the sample [m²]

Guarded hot plate

- placing a solid sample of fixed dimension between two temperature-controlled plates
- one plate is heated while the other plate is cooled, and their temperatures are monitored until they are constant



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erials	Material λ	[W.m ⁻¹ .K ⁻¹]	1
mat	• Copper	~370	
ing	Aluminium	~200	
nild	Carbon steel	~50	
a 8	Concrete	~1,4	
	• Glass	~0,75	
	• Brick	~0,7	
	• Water (20° C, quiet)	~0,60	
	• Wood	. ~0,15	S
	Mineral fibers		nal
	 Polystyrene foamed 	~0,035	nern ulat
	 Air (dry, quiet) 	0,025	thermal insulations
	Argon (quiet)	~0,015	

Thermal resistance R-value

$$\mathbf{R} = \mathbf{d} / \lambda \left[\left(\mathbf{m}^2 \cdot \mathbf{K} \right) / \mathbf{W} \right]$$

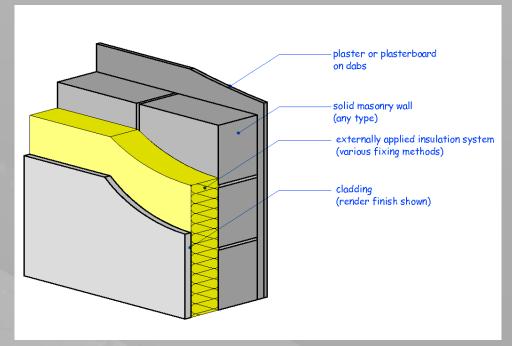
- directly proportional to the thickness of the material
- for construction, not material
- can be used for masonry blocks

 $R = 0.65 (m^2 \cdot K)/W$



Thermal resistance

multi-layered construction: the R-values of the individual layers are summed

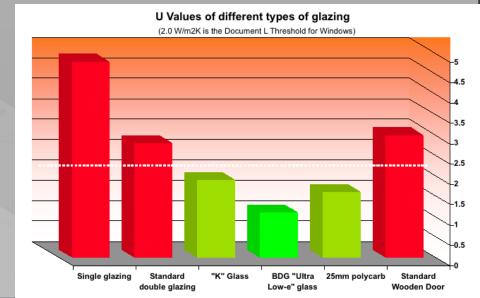


 $R_{total} = R_{outside air film} + R_{render} + R_{insulation} + R_{brick} + R_{plaster} + R_{inside air film}$

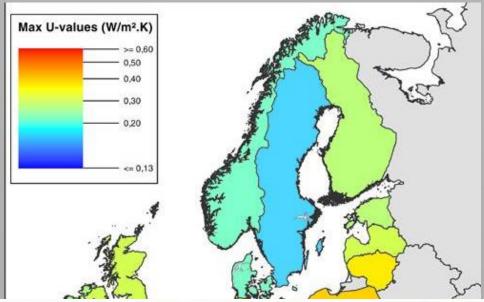
Thermal transmittance U-value

- the measure of the rate of heat loss through a material
- incorporates the thermal conductance of a structure along with heat transfer due to convection and radiation

U-value = 1/R[W / $m^2 \cdot K$]



U-value in Europe

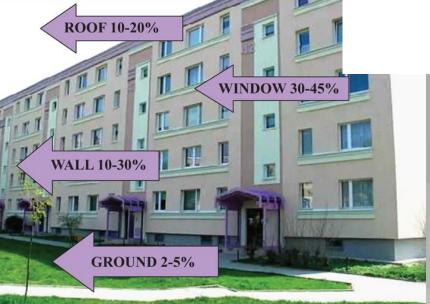


Rank	Country	Walls	Country	Roof	Country	Floor	Country	Windows
1	Sweden	0.3	Sweden	0.2	Sweden	0.2	Finland	1.9
2	Finland	0.4	Denmark	0.3	Denmark	0.4	Austria	2.3
3	Denmark	0.5	Finland	0.3	Finland	0.4	Denmark	2.4
4	Czech Republic	0.8	Czech Republic	0.6	Germany	0.8	Sweden	2.5
5	Austria	0.9	Austria	0.6	Czech Republic	0.9	Germany	2.7
6	Germany	0.9	Ireland	0.7	Belgium	0.9	Czech Republic	2.7
7	UK	1.0	Germany	0.7	France	1.0	France	3.1
8	Netherlands	1.1	UK	1.1	Ireland	1.0	Netherlands	3.2
9	France	1.2	Netherlands	1.2	Austria	1.0	Belgium	3.8
10	Ireland	1.2	France	1.3	UK	1.2	Ireland	3.8
11	Belgium	1.5	Belgium	1.6	Netherlands	1.3	UK	3.9

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Heat losses







Specific heat capacity c

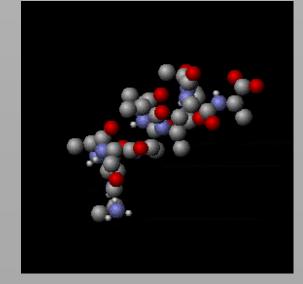
- = specific heat
- the amount of heat required to change
 1 kg of substance's temperature by a 1

K

units: [J.kg⁻¹.K⁻¹]

Depends on:

- temperature
- moisture: $c = c_0 + 0.42 \text{ w}_m$



Specific heat of some materials

Material	C [kJ.kg ⁻¹ .K ⁻¹]	
asphalt	0,92	
concrete	0,88	
brick	0,84	
glass	0,84	
copper	0,39	
granite	0,79	
gypsum	1,09	
water	4,18	
wood∥ to fibers	0,42	

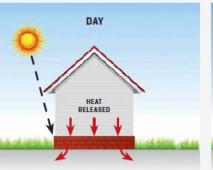


Heat accumulation

Thermal mass of building

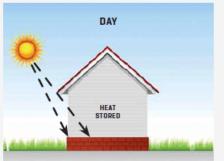
 absorbs thermal energy when the surroundings has higher temperature than the material and give thermal energy back when the surroundings are cooler

SUMMER SUN CONTROL AND THERMAL MASS





WINTER SUN CONTROL AND THERMAL MASS





Interior thermal control

Phase - change materials (PCM)

- materials with very high latent heat storage capacities
- best melting point for building purposes about 25 °C
- heat is absorbed or released when the material changes from solid to liquid and vice versa

Materials:

- paraffin
- fatty acids
- salt hydrates

Heat storage



Environmental temperature rises



Phase changing material becomes liquid



Managed temperature remains constant

Heat release



Environmental temperature falls



Phase changing material becomes solid



Managed temperature remains constant

Thermal effusivity

 determines the interfacial temperature when two semi-infinite objects at different temperatures touch

$$\mathbf{b} = \sqrt{\lambda \cdot \mathbf{c} \cdot \rho_{\mathbf{v}}}$$

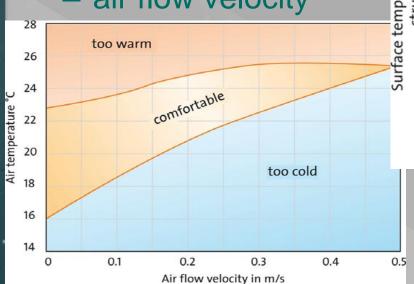
- units: [W.s^{0,5}.m⁻².K⁻¹]
- the higher b is, the colder sensation gives the material

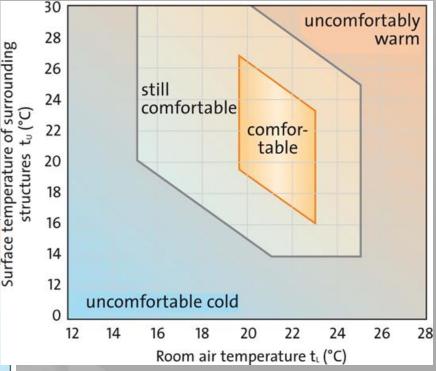


Thermal comfort

- the condition of mind which expresses satisfaction with the thermal environment
- affected by

 - air temperaturetemp. of surrounding surfaces
 - air flow velocity







Reflectivity

the ability of a surface to reflect radiation

 light reflectivity - the percentage light reflected from a surface at a given











Heat reflection and absorption

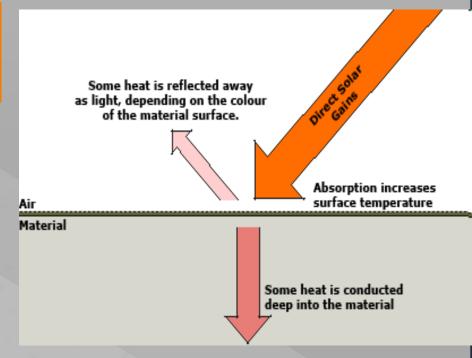
- reflection ρ
- absorption α

$$\alpha + \rho = 1$$

Heat reflectance:

$$R = \frac{\alpha}{\text{incidentheat}}$$

- black body R =1
- white R = 0.5



Solar absorptivity

Solar Absorptivity Chart for Selected Colours

Code	Colour	Absorptivity
6068	Black	0.94
6062	Dark Brown	0.91
6154	Metro Brown	0.89
6073	Dark Green	0.89
6072	Charcoal	0.89
6084	Navy Blue	0.87
6079	Heron Blue	0.85
6078	Green	0.84
6067	Slate Blue	0.8
6082	Regent Grey	0.75
6071	Stone Grey	0.6





Faculty of Civil Engineering

Heat reflective paints

