



Building Materials

Lecture 12



Bituminous materials





Bituminous materials

- an amorphous, black or dark-colored, (solid, semi-solid, or viscous) cementitious substance, composed principally of high molecular weight hydrocarbons
 - **asphalts**
 - **tars**
 - **nonvolatile**
 - **nontoxic**
 - **soften when heated**
 - **soluble in carbon disulfide**



Babylonian tablet of the period of Agade, circa 2700 B.C. A letter concerning the receipt of bitumen



Bituminous materials

- asphalt
 - natural
 - from petroleum processing
- tar
 - do not occur in nature
 - condensates in the processing of coal, petroleum, oil-shale, wood or other organic materials
- pitch
 - formed when a tar is partially distilled so that the volatile constituents have evaporated off from it



Natural asphalt



Gilsonite (natural asphalt) mine
Iran





Petroleum asphalts



- refined residue from the fractional distillation of crude oils = **primary asphalt**
 - **oxidised a.** - passing air through asphalt at elevated temperature
 - **less sensitive to thermal changes**
 - **modified a.** - their properties (elasticity, adhesive or cohesive strength) have been modified by the addition of polymer
 - **SBS** (styren-butadien-styren, 7-15%)
 - **higher ductility** (several hundred %)
 - **APP** (atactic polypropylene, 15-35 %)
 - **higher resistance against UV radiation**
 - **better adhesion**



Asphalt properties

General:

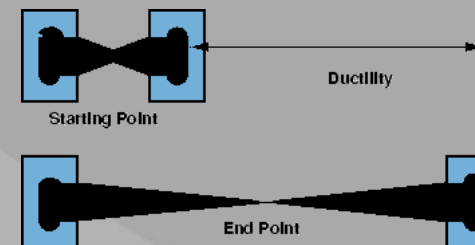
- insoluble in water
- no absorptivity
- density $\approx 1000 \text{ kg.m}^{-3}$
- $\lambda \approx 0,75 \text{ w.m}^{-1}.\text{K}^{-1}$
- soluble in organic solvents
- thermal elongation $\approx 600.10^{-6} \text{ K}^{-1}$
- combustible
- ageing by UV radiation and O_2





Influence of modification on properties

Property Asfalt type	Breaking point [°C]	Thermal stability [°C]	Softening point [°C]	Ductility [%]
oxidized	0 – 4	70	85 – 90	2 – 5
APP modification	-5 až -15	115 – 130	135 – 150	20
SBS modification	-15 až - 35	90 – 110	110 – 125	> 100
SBS-SIS-SEBS modification	- 45	105	125	





Asphalts testing

Special tests:

- penetration
- softening point
- breaking point
- ductility
- viscosity
- coating ability





Penetration test

- the depth to which a needle penetrated an asphalt sample under specified conditions of load, time and temperature
- units: 0,1 mm



Procedure





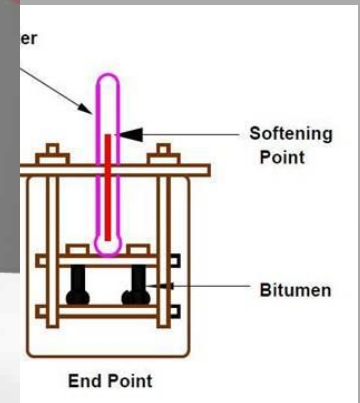
Softening point test

- Ring-and-ball method
- the softening point is the temperature at which a material softens beyond some arbitrary softness

– heating by $5^{\circ}\text{C}/\text{min}$



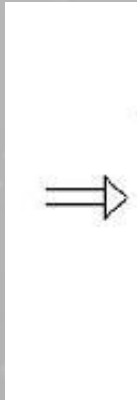
Temperature, specimen (mm)





Breaking point test

- **Fraass method**
- to determine the temperature below which a bitumen tends to break rather than to flow when stressed



the





Ductility test

- stretching a standard-sized briquette of asphalt binder at standard temperature to its breaking point
 - **ductility** = stretched distance in cm at breaking



DUCTILITY TEST

The ductility value of bitumen binder is expressed as the distance in centimeters to which a standard briquette of bitumen can be elongated before the thread ruptures.





Asphalt products

- paints
- mastic
- **asphalt cement**
 - asphalt concrete
- emulsions
- suspensions
- **membranes (felts)**
- roofing





Asphalt waterproof membranes

release film

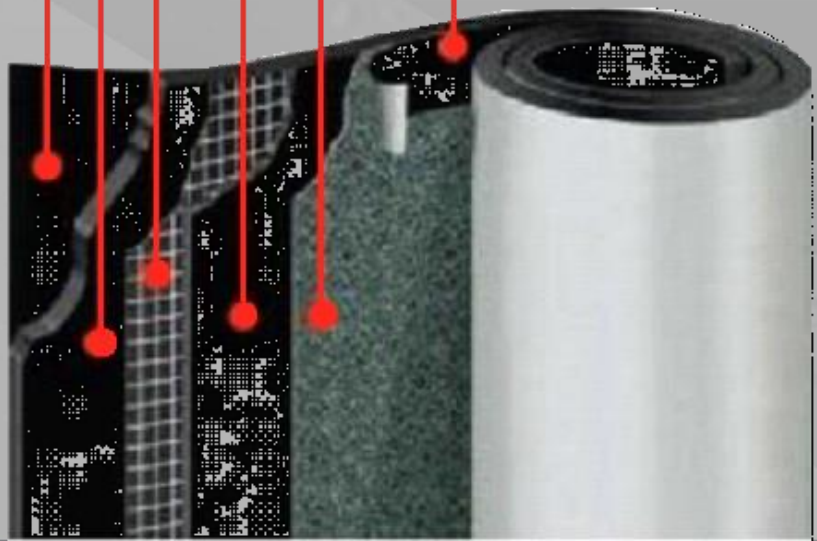
bitumen layer

base (PE, glass-fibre, copper)

bitumen layer

mineral granules (sand, slate)

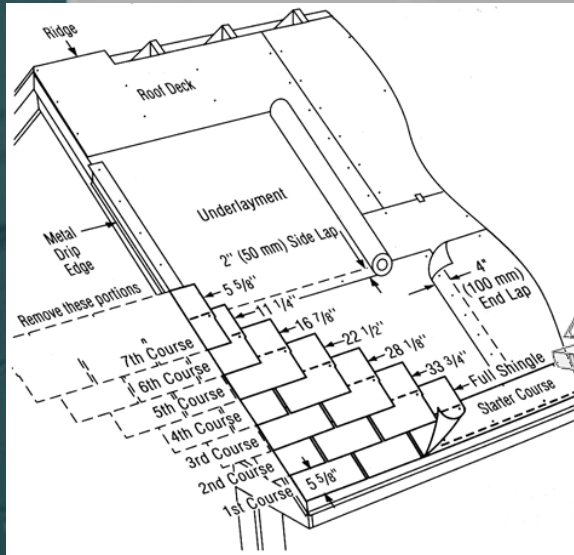
protective strip





Asphalt shingles

- roof covering
- cutted from asphalt membranes
- wood roof decking
- fixation by g





Asphalt shingles - roof

- minimal **roof slope** - 15°
- **square weight** - 9,5 to 14 kg/m²
- **life expectancy** - 30 to 50 years
- + lightweight
- + price
- lower fire resistance
- softening at higher temperatures
- high demands on realization
- shorter durability
- decking under shingles





Asphalt paints

- asphalt + organic solvents
 - cold applied - primers, insulating coats, reflex paints, waterproof varnish, mastic (+aggregates)
 - hot applied – grouts, sealants, asphalt concrete
- asphalt + water + emulsifying agent
 - emulsions
 - suspension (+ filler)
 - waterproofing layers, joining, grouts, sealants





Asphalt concrete



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Building materials



POLYMERS



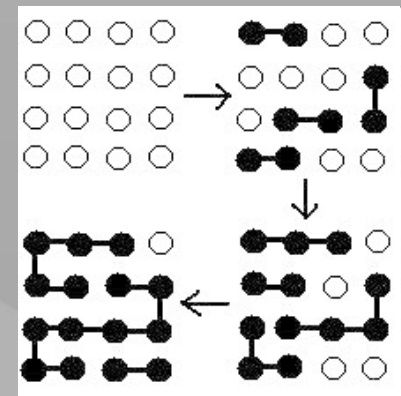


Polymers = plastics

- *polys = many, meros = parts*
- a large molecule (macromolecule) composed of repeating structural units (100 -100 000)



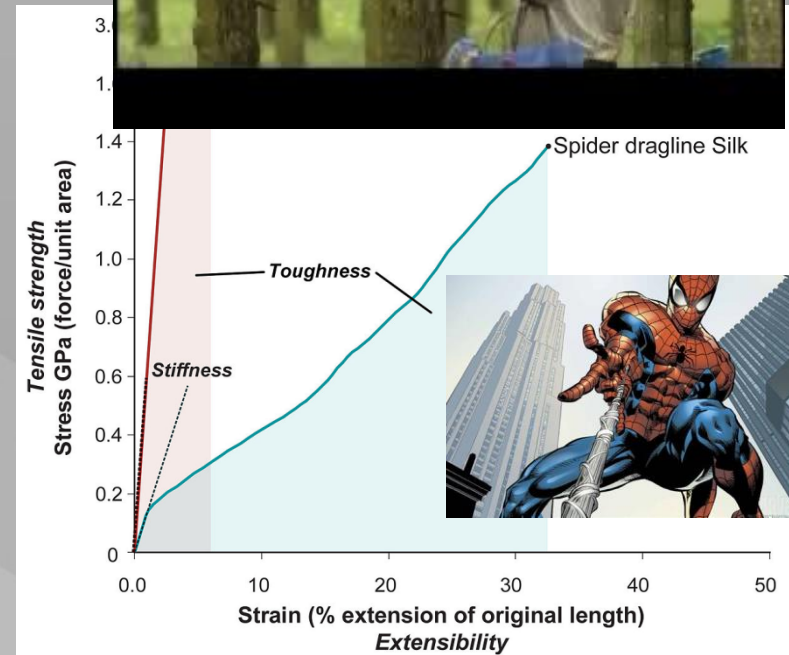
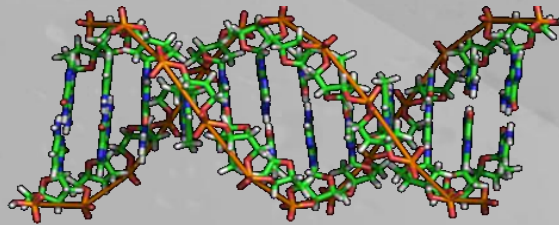
- **polymerization** - a process of reacting monomer molecules together in a chemical reaction to form polymer chains or three-dimensional networks
 - polycondensation
 - polyaddition
 - radical polymerization





Natural polymers

- rubber – rubber tree (*Hevea brasiliensis*)
- cellulose – cotton, wood – viscose, cellophane
- starch
- proteins – keratin, collagen, casein, DNA





Polymer abbreviations

- **CE** cellulose / cellophane
- **EP** epoxy
- **MF** melamine formaldehyde
- **PA** polyamide
- **PE** polyethylene
- **PET** polyethylene terephthalate
- **PF** phenolformaldehyde
- **PP** polypropylene
- **PS** polystyrene
- **PU,PUR** polyurethane
- **PVC** polyvinyl chloride



PETE



HDPE



V



LDPE



PP



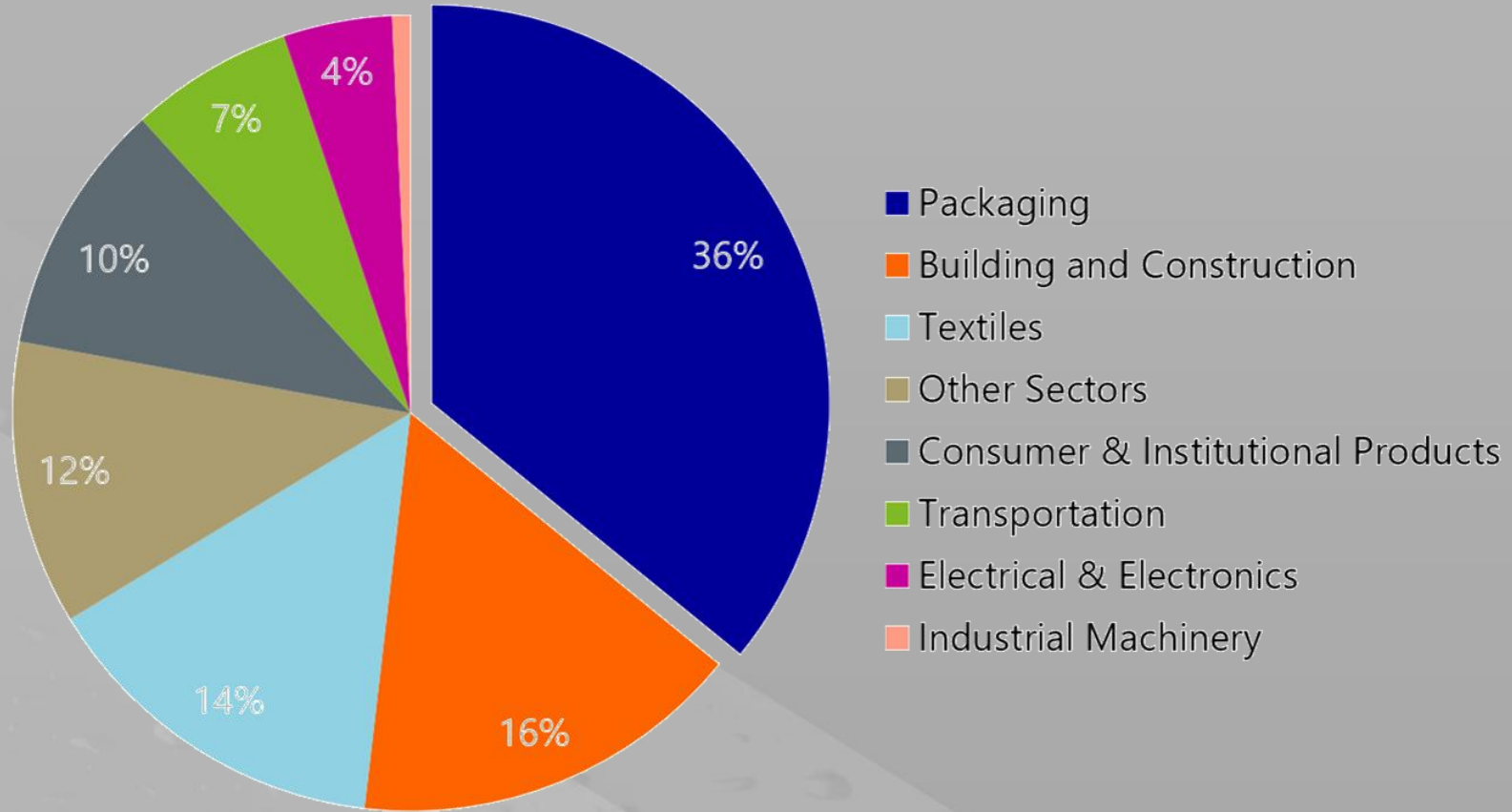
PS



OTHER



Polymer consumption





Polymers

Thermoplastics

- turns to a liquid when heated and freezes to a rigid state when cooled sufficiently
- can be remelted and remoulded
- **PE, PVC, PS**



Thermosets

- undergo an irreversible chemical change by curing
- the cure may be done through heat, a chemical reaction or irradiation
- molecular chains are cross-linked into rigid 3-D structure
- **PF, PU, EP**



Polymers - mechanical properties

Polymer	Density (kg/m ³)	Tensile Strength (N/mm ²)	Elongation (%)	Young's Modulus (GN/m ²)	Brinell Hardness Number
PVC	1330	48	200	3.4	20
Polystyrene	1050	48	3	3.4	25
PTFE	2100	13	100	0.3	
Polypropylene	900	27	200-700	1.3	10
Nylon	1160	60	90	2.4	10
Cellulose Nitrate	1350	48	40	1.4	10
Cellulose Acetate	1300	40	10-60	1.4	12
Acrylic (metacrylate)	1190	74	6	3.0	34
Polyethylene	950	20-30	20-100	0.7	2
Epoxy resin, glass filled	1600-2000	68-200	4	20	38
Melamine formaldehyde, fabric filled	1800-2000	60-90		7	38
Urea formaldehyde, cellulose filled	1500	38-90	1	7-10	51
Phenol formaldehyde, mica filled	1600-1900	38-50	0.5	17-35	36
Acetals, glass filled	1600	58-75	2-7	7	27

thermoplastics

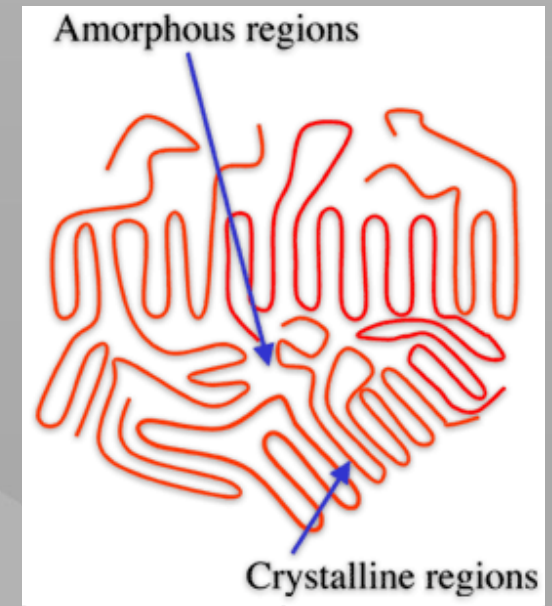
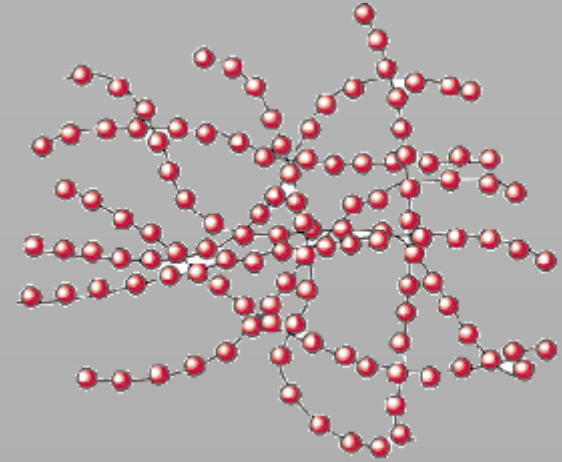
thermosets





Thermoplastics

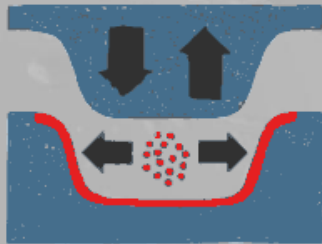
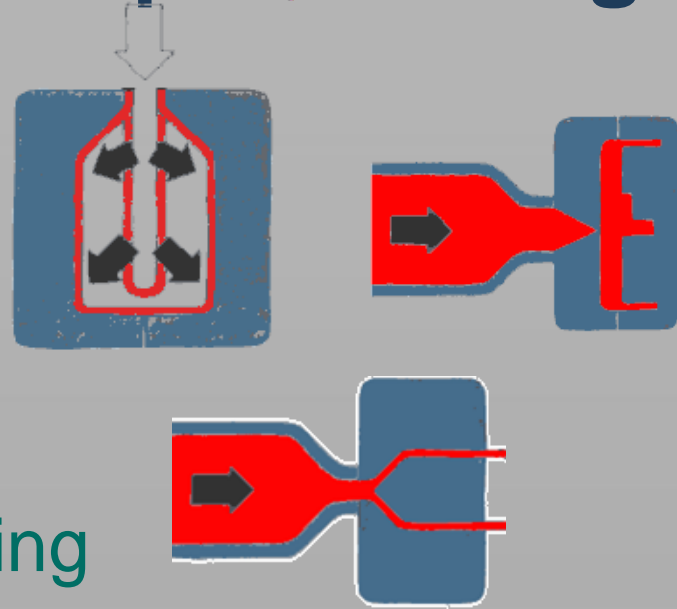
- consist of long molecular chains with no (or very little) regular structure
- chains are tangled (Brownian motion)
- **semi-cristalline regions** - a regions in a plastic in which the molecule arrangement is ordered and structured
 - increase the melting point of the plastic (more heat is needed to encourage molecular movement)





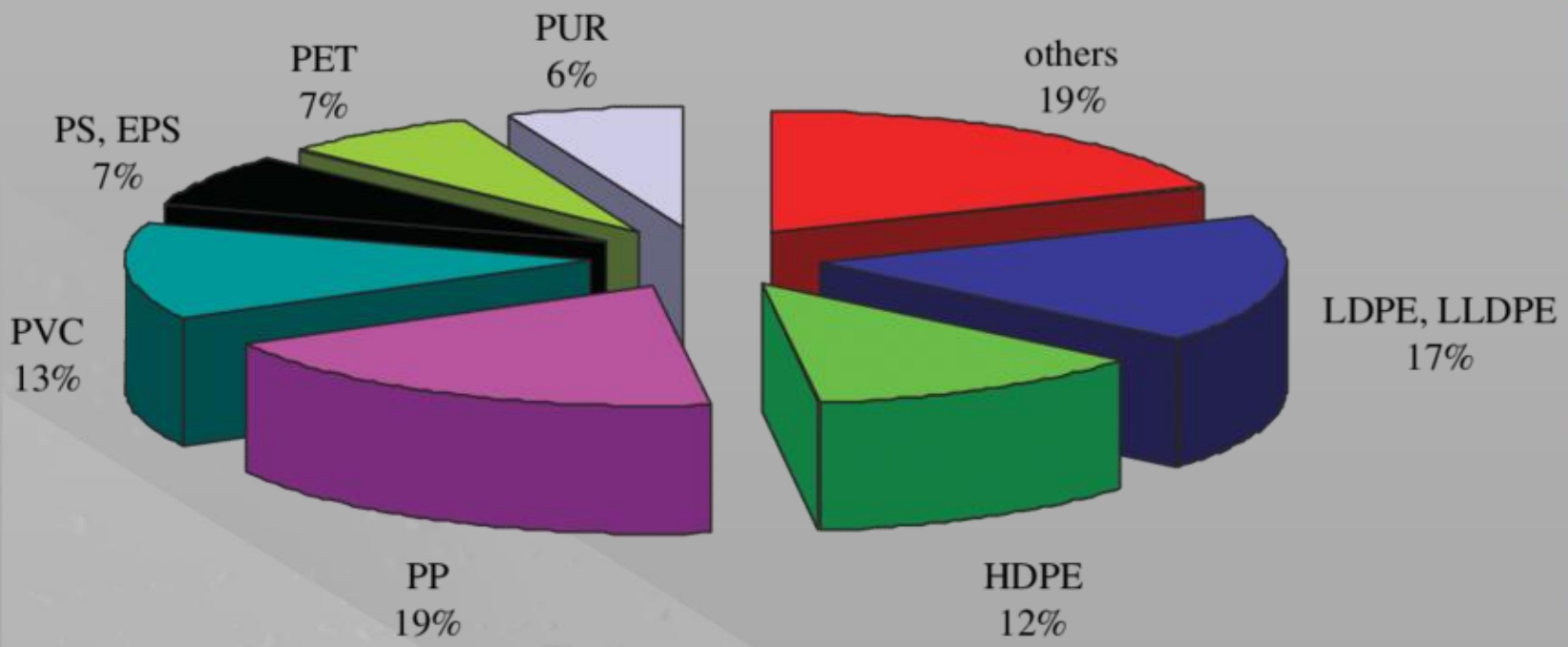
Thermoplastic processing

- pellets or powder
 - extrusion
 - blow molding
 - injection molding
 - compression forming
 - thermoforming
 - calendering (rolling)





Polymer consumption according type

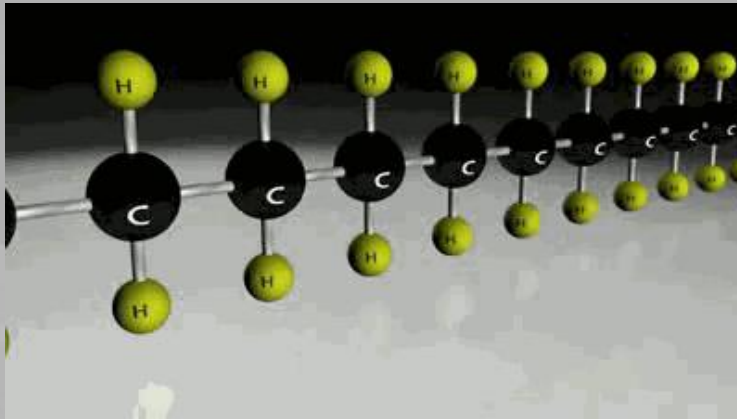


Europe, 2013



Polyethylene PE

- most common plastics (80 mil. tons /year)
- polymerization from ethylene

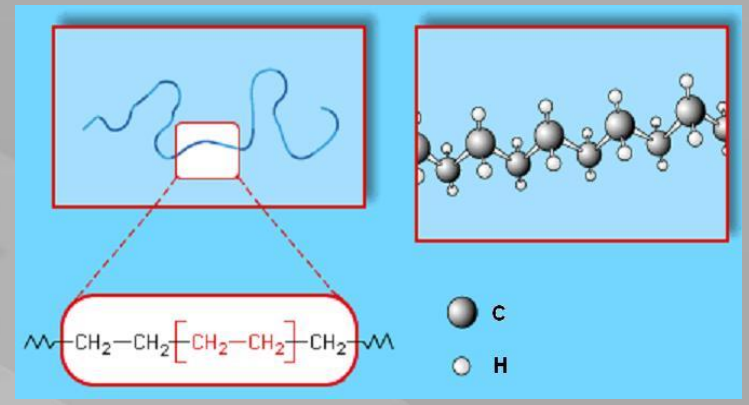
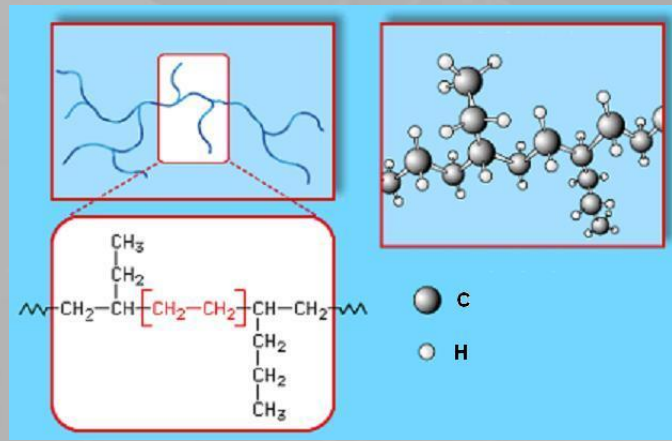


- low-density polyethylene - **LDPE**
- high-density polyethylene - **HDPE**



LDPE + HDPE

- **LDPE - low density** (density 0,910– 0,940 g.cm³)
 - a high degree of short and long chain branching
 - chains do not pack into the crystal structure
- **HDPE - high density** ($\geq 0,940$ g/cm³)
 - a low degree of branching
 - stronger intermolecular forces and tensile strength





Polyethylene



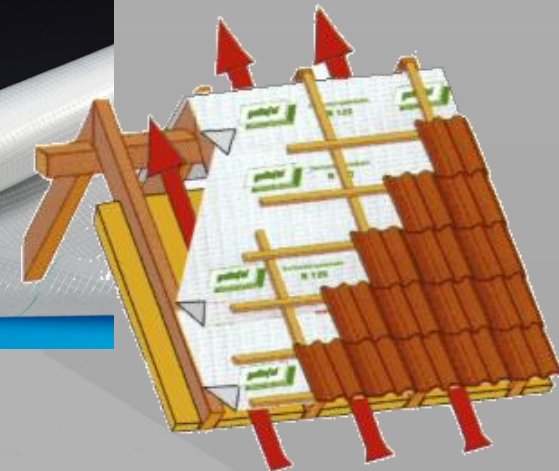
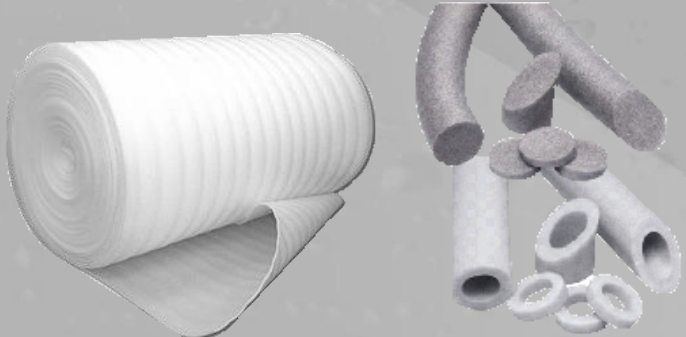
- + **pliable** (in thin layer)
- + good **shape memory** (LPDE)
- + high **strength** and **thermal resistance** (HDPE)
 - utilization above 100 °C
- + excellent **chemical resistance** at normal temperature
 - strong acids and alkalis, HF!
- + **no health risks**
- + good **recyclability**
- **difficult gluing** → joining by hot air welding
- **dissolution** in organic solvents at elevated temperatures
- **diffusion of some materials** into PE (fats, oils)
- **low weather resistance** – UV stabilization
- **flammable, drips** at burning



LDPE

LDPE products

- stretch and bubble wraps
- vapor permeable roof foils
- corrosion inhibiting foils
- waterproof insulation of tunnels (better fire behavior than HDPE)
- expanded foam – thermal insulations
 - closed cell,
 - $\lambda \approx 0,045 \text{ W.m}^{-1}.\text{K}^{-1}$





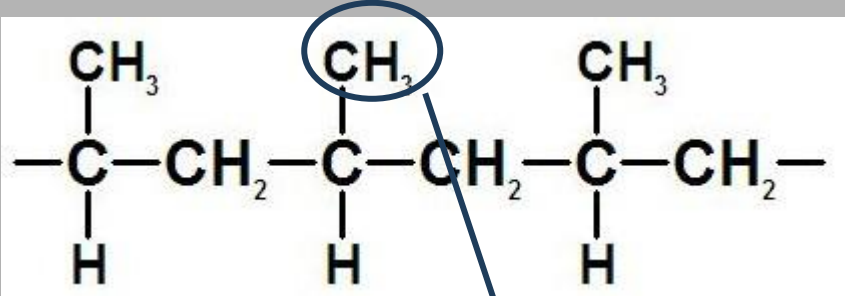
HDPE products

- membranes – smooth, textured, drainage, radon-proof
- sheets – corrosion resistant wall coverings, radiation shielding,
- pipes – water, chemical resistant
- wood-plastic composites





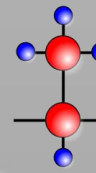
Polypropylene PP



methyl

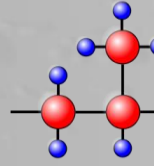


- isotactic



- syndiotactic

- atactic

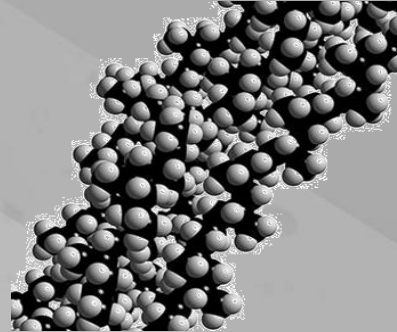
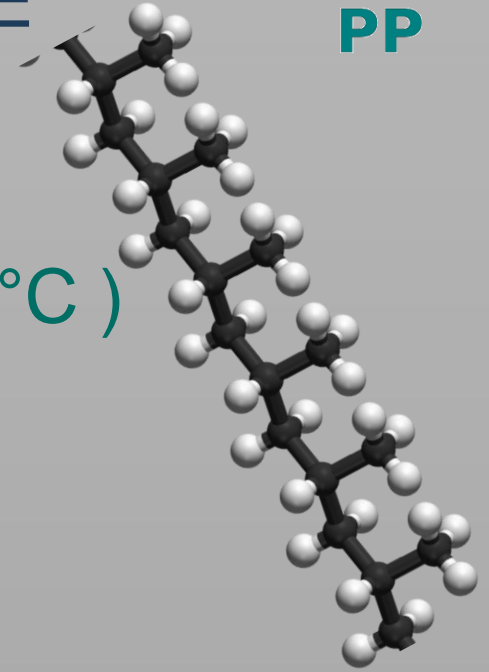




PP properties



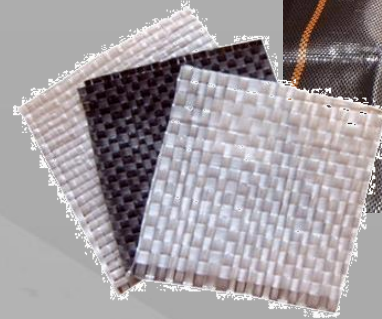
- **isotactic PP** – similar to HDPE
 - higher strength
 - good resistance to abrasion
 - higher melting point ($\sim 165\text{ }^{\circ}\text{C}$)
 - brittle at temp. $< 0^{\circ}\text{C}$ (modification by rubbers)
 - low weather resistance
- **atactic PP**
 - amorphous
 - rubbery
 - low strength





PP products

- tubes and pipes
- sheets, boards
- fibers
 - concrete, plasters
 - textiles (woven and non-woven)
- geomembranes
- **APP** – asphalt modification, additive to adhesives





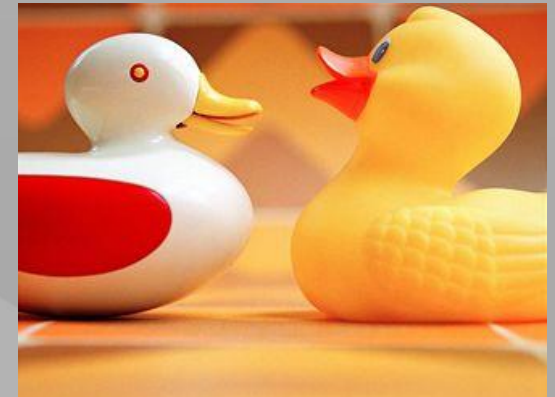
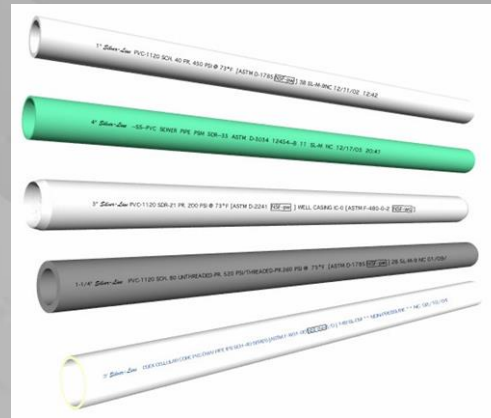
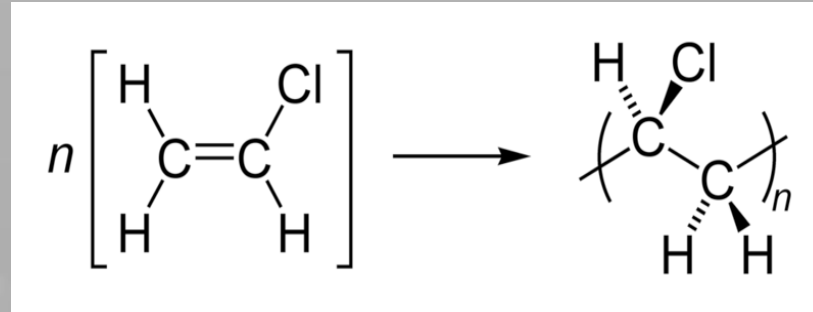
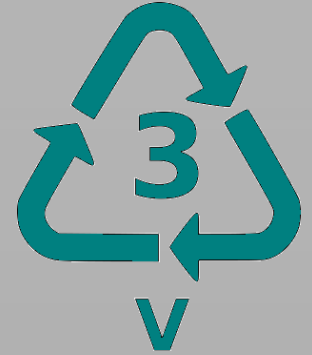
PP fibres in concrete

- 0,9 kg /1 m³ of concrete
 - 150 millions of fibres
- surface treatment
 - easy dispersion in mixture
- mixing in normal mixer
- < shrinkage and cracking
- > impact and shatter resistance
- > resistance to frost
- > durability
- < spalling (at fire)



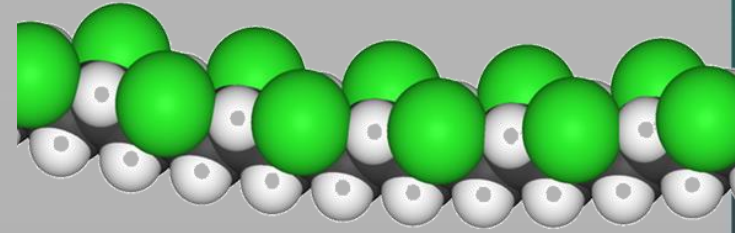


Polyvinylchloride PVC





PVC properties



- density: 1380 kg/m^3
- boiling point: $212 \text{ }^\circ\text{C}$
- pure PVC – rigid and brittle → plasticizing
- temperature range: -25°C to $+70^\circ\text{C}$
- thermally sensitive – heat stabilizers
- self-extinguishing
- glueable
- production of toxic gases, when burning (dioxins, hydrochloric gas)
- difficult to recycle



PVC plasticizing



Plasticizers:

- phthalates (CMR ? = carcinogenic, mutagenic or toxic to reproduction), adipates, citrates
- **rigid PVC**
 - less than 12 % plasticizers
- **flexible PVC**
 - 20 - 40 % plasticizers
- **unplasticized PVC (uPVC)**
 - only minimal amount of plasticizers necessary for production - “vinyl”



Waldo Semon
(1898-1999)



PVC products

- hard
 - sheets, boards, pipes and fittings, vinyl siding
- soft
 - waterproof membranes, insulation on electric cables, flooring, coating on steel





PVC suspended stretch ceiling

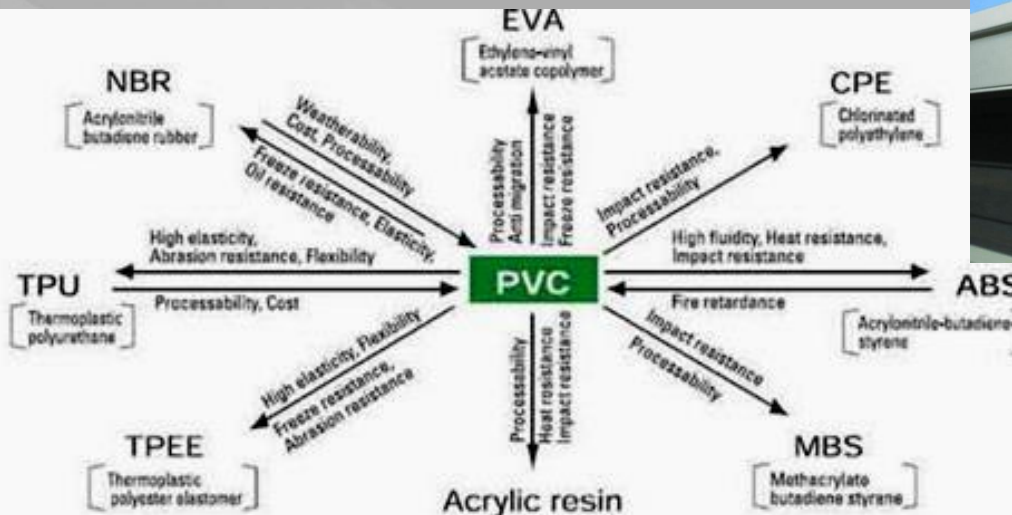
- biaxially stretched film
- at instalation heating to 60 °C
- stretches after cooling
→ perfectly smooth surface





PVC copolymers

- chlorinated **PVC** (CPVC)
 - addition of chlorinated polyethylene → weather resistance
 - window frames, gutters, facade boards



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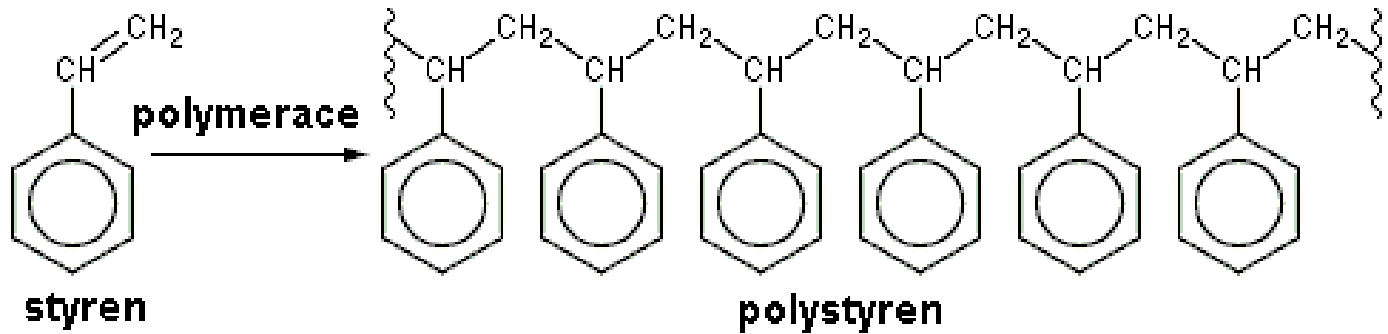
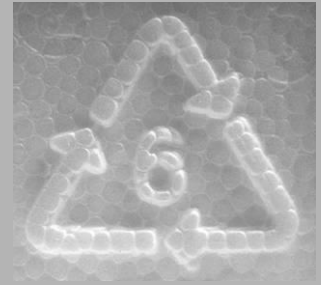
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Building materials



Polystyrene PS



- hard and brittle
- mostly used foamed
 - **expanded EPS**
 - **extruded XPS**

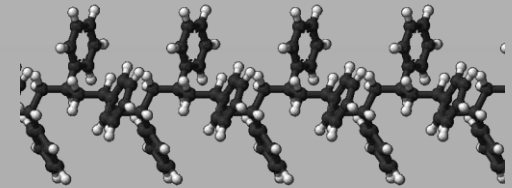




Expanded polystyrene EPS

Production:

- styrene is made by combining ethylene and benzene
- suspension polymerization → raw polystyrene beads (\varnothing 1 mm)
- heating to 100°C and blowing by pentane → pre-expanded beads
- after several days final expansion into blocks



Raw Beads

Expanded Beads



EPS production





Expanded polystyrene EPS

Properties:

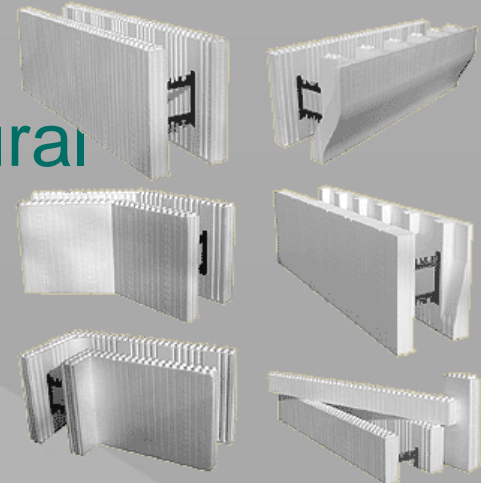
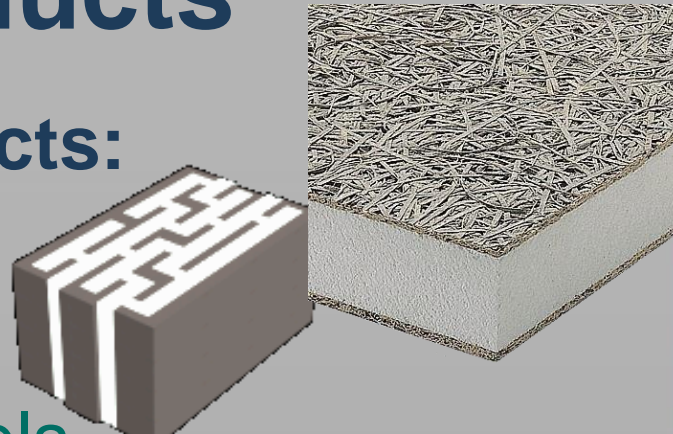
- $\rho_V = 5-100 \text{ kg.m}^{-3}$ (20, 25 a 35 kg.m^{-3})
- $\lambda \approx 0,04 \text{ W.m}^{-1}\text{K}^{-1}$
- **flammable** → addition of retarder - self extinguish
- **volume changes** – off-gassing of pentane, thermal elongation
- water **sorptivity** 2 - 5 % (volumetric)
- water vapor diffusion resistance $\approx 30-70$
- easy cutting and gluing



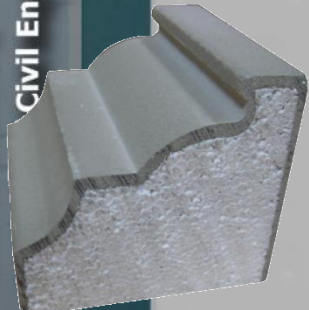
EPS - products

- thermal insulating products:

- blocks and boards
- pipes insulations
- structural insulated panels (sandwich + gypsum board, fiberboard, asphalt felts)
- insulating concrete forms
- loose beads
- non-weight-bearing architectural elements



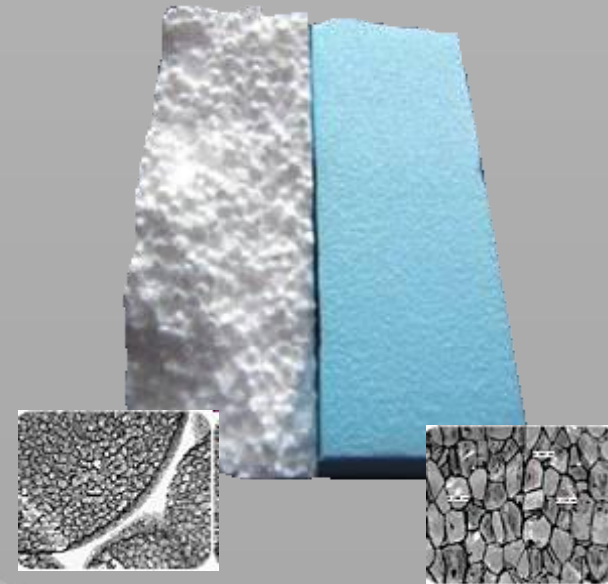
- ornamental molding
- ceiling tiles,





Extruded polystyrene XPS

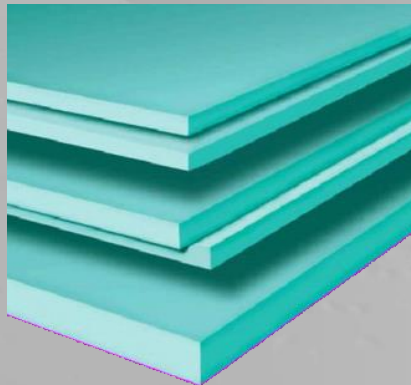
- **polystyrene beads + additives** are melted in an extruder and a **blowing agent** is injected into the extruder under high pressure where it dissolves into the polystyrene melt
 - **closed cells**
- **properties**
 - $\rho_v = 25, 40, 45 \text{ kg.m}^{-3}$
 - $\lambda \approx 0,025 \text{ W.m}^{-1}\text{K}^{-1}$
 - lower sorptivity (0,2%)
 - higher strength





XPS - products

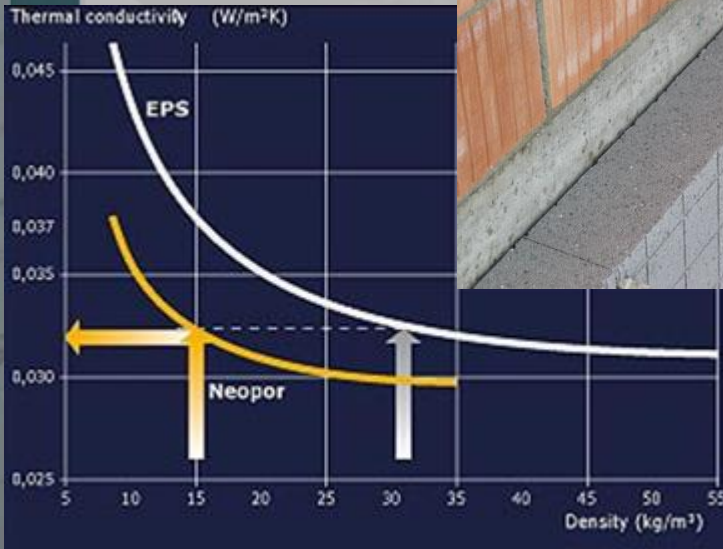
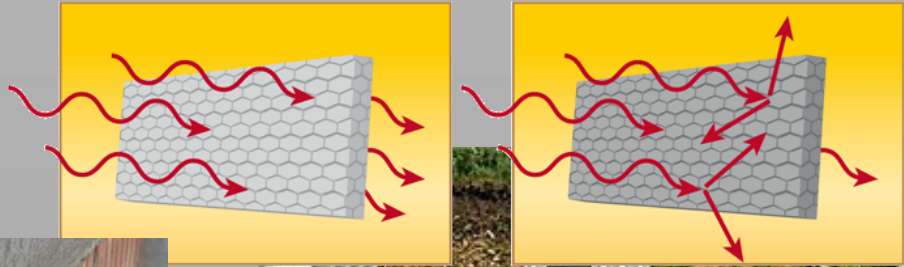
- similar to EPS
- where lower sorptivity and higher strength is necessary
 - roofs (green roofs)
 - under tiles
- usually colored





Grey polystyrene - Neopor

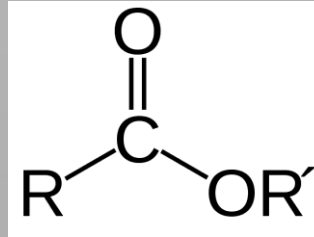
- adding of fine graphite powder \rightarrow lower thermal radiation through material
- lower λ at lower ρ_v





Polyesters

- contain the ester functional group in their main chain
- fibers



- **PET** (polyethylene terephthalate)
 - bottles
 - cloth
 - geomembranes
 - recycled PET





Recycled PET

- membranes
- geotextiles
- fence elements
- paving elements
- boards

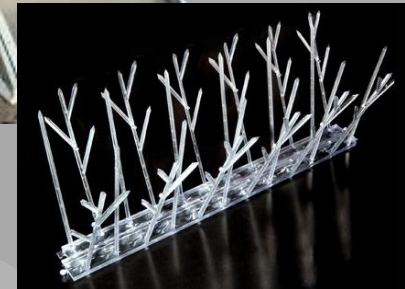
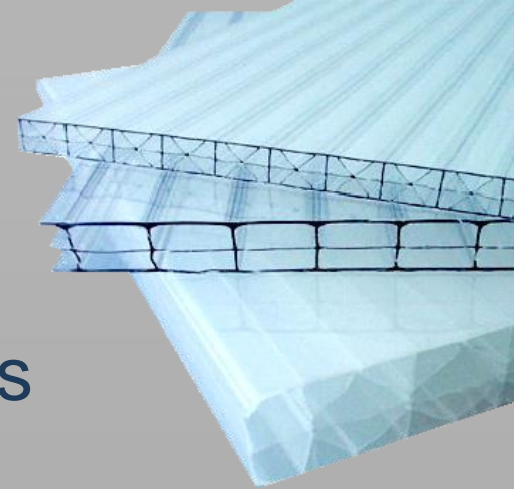




Polyesters

Polycarbonates PC

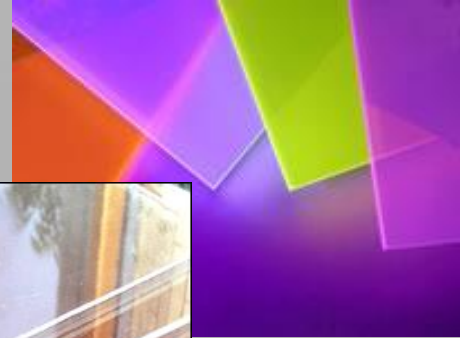
- transparent, colorless
 - good mechanical properties
 - weather resistance
 - relatively small volume changes
- replacement of glass





Polyesters - - polymethylmetacrylate PMMA

- acrylic glass, „Plexiglass“
- density $\sim 1200 \text{ kg.m}^{-3}$
- low surface hardness
- good impact strength
- transparent
- relatively high thermal expansion
 - heat stabilization
 - a lightweight and shatter-resistant alternative to glass



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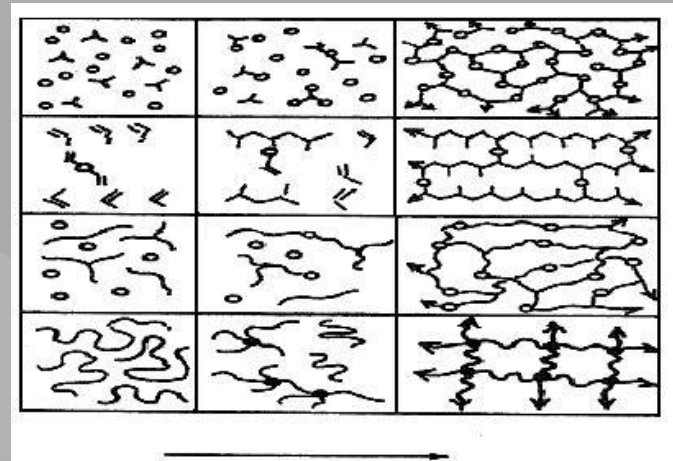
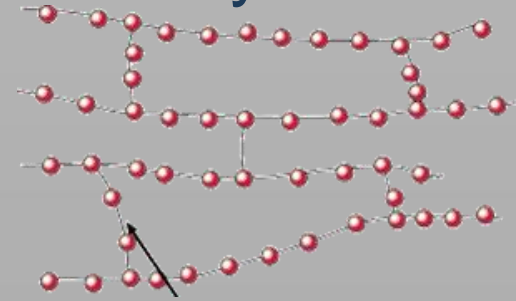


Building materials



Thermosets

- the polymer chains are cross-linked by intermolecular bonding into 3-D structure
- after curing are irreversibly changed (hardened) and cannot be melted
- curing – heat, chemical reaction, irradiation
- **cured** – solid products
- **uncured** - adhesives, paints, glues





Formaldehyde resins

- phenol formaldehyde resin
 - thermal insulating boards and laminates from glass or mineral fibres
 - boards with wood flour (Bakelite)
 - binding agent in acid-resistant mortars
 - released formaldehyde
- urea-formaldehyde is not used now – health risks

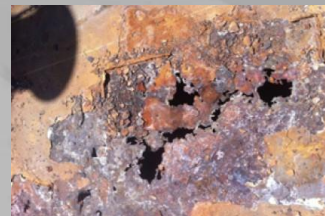


L. H. Baekeland
(1863-1944)



Fenolic foam

- $\rho_V = 30 - 80 \text{ kg.m}^{-3}$
- $\lambda \approx 0,04 \text{ W.m}^{-1}\text{K}^{-1}$
- low strength
- good thermal and fire resistance
- low price
- caused corrosion on corrugated steel roofs

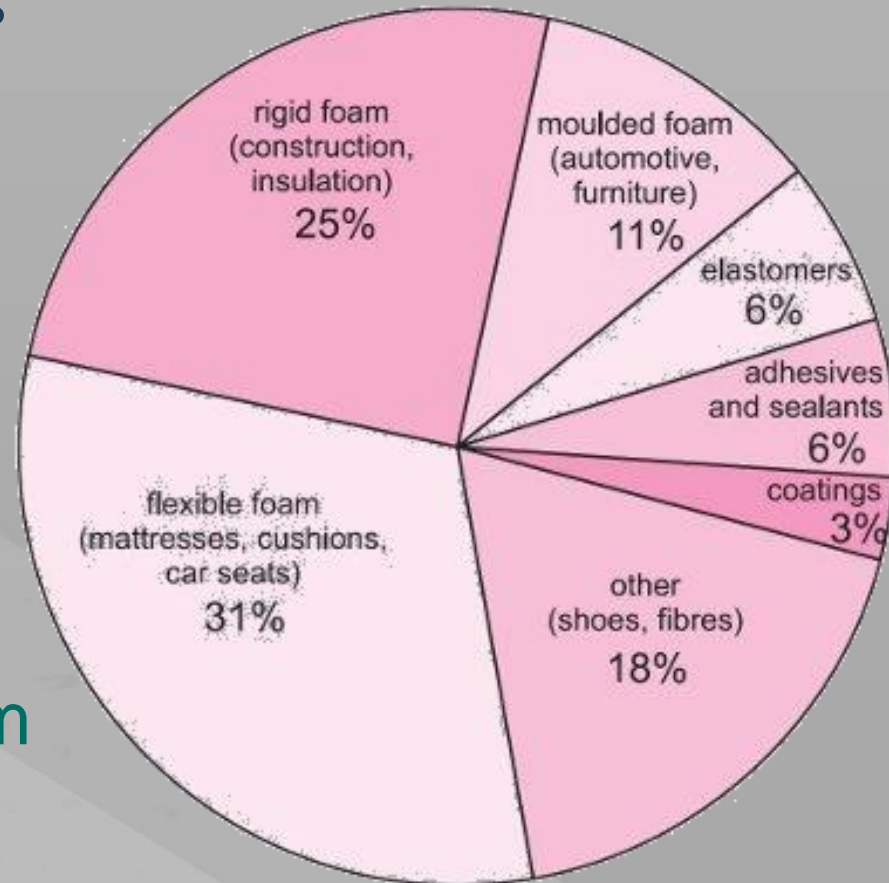




Polyurethanes PUR

- reaction of a polyol with a diisocyanate or a polymeric isocyanate in the presence of suitable catalysts and additives

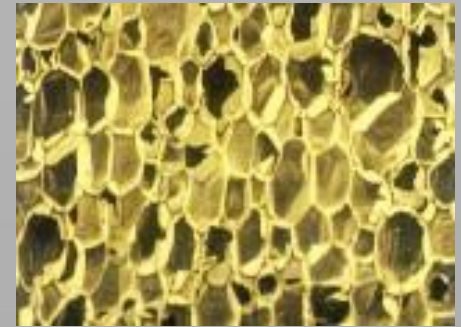
- coatings
- adhesives
- sealants
- binder
- rigid or flexible foam





Rigid PUR foam

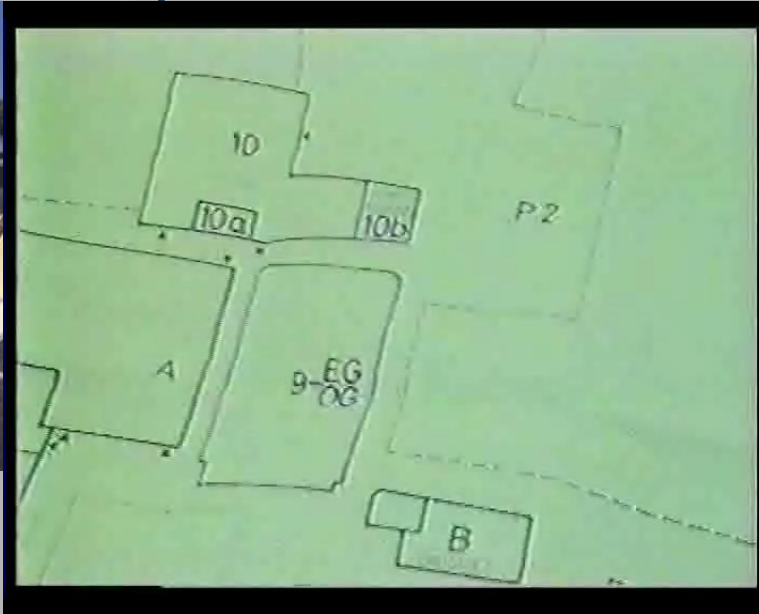
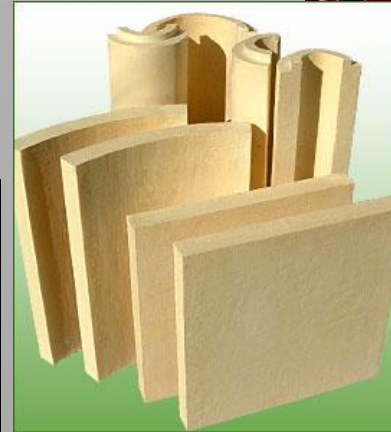
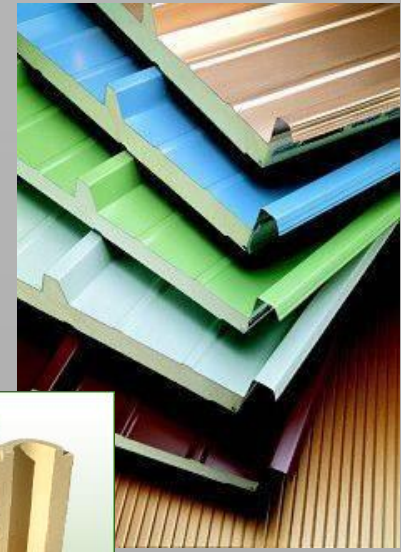
- two components
- high strength
- $\rho_V = 28 - 34 \text{ kg/m}^3$
- $\lambda = 0,03 \text{ W.m}^{-1}\text{K}^{-1}$
- closed cell – low sorptivity - waterproofing
- low vapor diffusivity
- not UV resistant





Rigid PUR foam products

- thermal and waterproof insulation
 - **sprayed insulation**
 - need protective coating against UV radiation
 - **structural panels** and composites





PU montage foam

- for montage only!
- lower strength and durability
- open pores





Epoxy resins

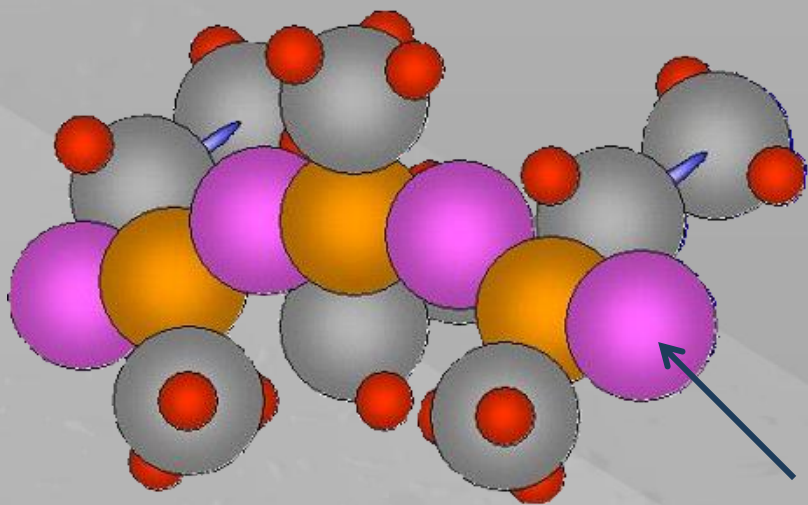
- reaction of an epoxy resin with polyamine hardener
- excellent adhesion to substrate
- large range of temperatures
- hardened product - high chemical and mechanical resistance
- low volume changes
- paints and coatings
- adhesives
- polymer concretes
- flooring





Silicones

- $[R_2SiO]_n$
- rubbers
- resins



siloxan chain
(...-Si-O-Si-O-Si-O-...)



Silicones - products

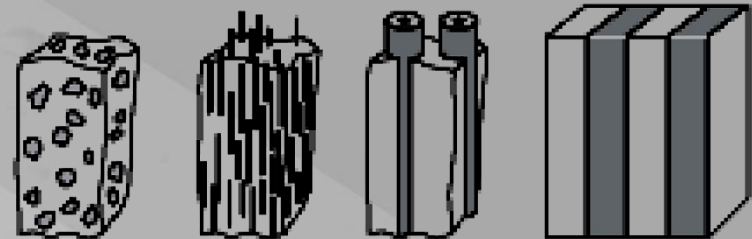
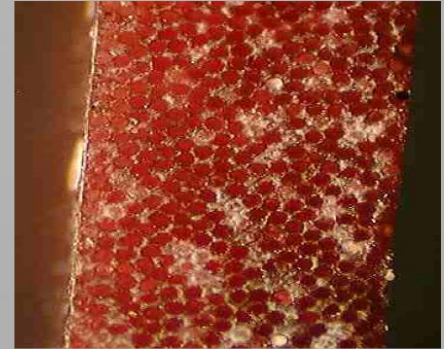
- „one-component“ caulks (hardening by air moisture)
 - very good adhesivity
 - some caulks react acidly → not suitable for concretes, mortars
 - different chemical resistance
- two-components rubbers – more liquid
- silicon gaskets
- hydrophobization (plasters, mortars, paints)





Composites with thermosets

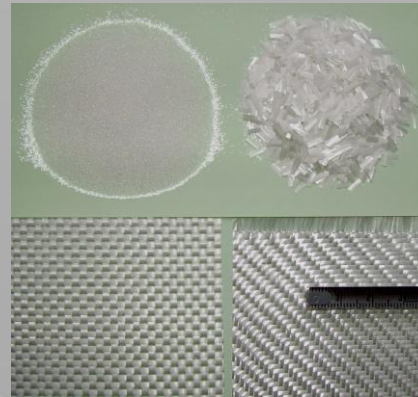
- **polymer matrix (binder)**
 - polyester
 - epoxy
 - vinyl ester
 - phenolic
- **reinforcement**
 - granular (particles)
 - short fibers (glass, carbon, aramid ...)
 - continuous fibers
 - laminate





Glass reinforced composites

- glass fibres, woven and nonwoven fabric, (50 - 80% of glass), microspheres



transmission tower
Ještěd, CR



Polymer concretes

- use polymers to supplement or replace cement as a binder
 - + high strengths
 - + rapid curing
 - + good adhesion to most surfaces
 - + good chemical resistance
 - + lightweight
- worse fire resistance
- price
- high demands on quality of mixing





Polymer concretes - properties

		Polymer concrete	Concrete
bulk density	kg/m ³	1 900 - 2400	2500-2700
modulus of elasticity	GPa	10 - 40	15 - 40
tensile strength	MPa	4 - 25	1 - 5
compressive strength	MPa	50 - 210	6 - 60
thermal conductivity	W/mK	1,5 - 2	1,28 - 1,54
thermal elongation	10 ⁶ K ⁻¹	10 - 50	9 - 12
mass sorptivity	%	0,02 – 1	6 - 13

